# Parasite glycoconjugates. Part 4. ${ }^{1}$ Chemical synthesis of disaccharide and phosphorylated oligosaccharide fragments of Leishmania donovani antigenic lipophosphoglycan 

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#### Abstract

The disaccharide 1, tetraglycosyl monophosphate 2, hexaglycosyl diphosphate 3 and octaglycosyl triphosphate 4, which are fragments of the phosphoglycan portion of Leishmania donovani lipophosphoglycan, have been synthesized. Elongation of the chain was performed using the suitably protected glycobiosyl hydrogenphosphonate derivatives 5 and 6 for the successive introduction of glycobiosyl phosphate units.


## Introduction

The Leishmania are trypanosomatid protozoan parasites that cause a variety of diseases in the tropics and sub-tropics. The geographical distribution and pathology of leishmaniasis varies according to the species of the Leishmania parasite. For example, Leishmania donovani causes visceral leishmaniasis (known as kalazar), characterised by an enlarged liver and spleen, that is often fatal, whereas $L$. major generally causes a self-limiting skin lesion (known as oriental sore). The parasites undergo a complex life-cycle between their insect vectors and their mammalian hosts. In the sand-fly, the parasites divide in the mid-gut as non-infectious procyclic promastigote forms and many of these parasites are found attached to the gut epithelium. After a few days, the parasites differentiate into non-dividing, infectious metacyclic promastigote forms that detach from the gut epithelium and migrate to the mouth parts. The metacyclic promastigotes are transmitted to a mammalian host in the sand-fly saliva during a blood meal, where they bind to and invade host macrophages to initiate the infection. They then rapidly differentiate into round amastigote forms of the parasite that undergo division inside the macrophage phagolysosome. Heavily parasitised macrophages eventually lyse and the released amastigotes invade adjacent macrophages to propagate the infection. Amastigotes in infected macrophages that are ingested by a sand-fly differentiate into procyclic promastigotes in the sand-fly mid-gut, thus completing the cycle.

The most abundant macromolecule on the surface of promastigote forms of all Leishmania species is a complex glycoconjugate called lipophosphoglycan (LPG). The general structure of $L P{ }^{2}$ is:

$$
\begin{gathered}
(\mathrm{R} \rightarrow 3) \\
\alpha \text {-D-Man } p-(1 \rightarrow 2)-\alpha \text {-D-Manp-(1-PO } \mathrm{Pa}_{3} \mathrm{H}-[-6)-\beta-\mathrm{D}-\mathrm{Gal} p-(1 \rightarrow 4)- \\
\alpha-\mathrm{D}-\mathrm{Man} p-\left(1-\mathrm{PO}_{3} \mathrm{H}-\right]_{n}-\text { glycosyl phosphatidylinositol } \\
\text { anchor, }
\end{gathered}
$$

where $n=14-30$. The nature of the R group substituting the 3 -position of the $\beta-\mathrm{D}-\mathrm{Gal} p$ residue of the common $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranosyl phosphate repeat unit varies according to the species of Leishmania. For example, in $L$. donovani the $\mathbf{R}$ group is simply $\mathbf{H}$, whereas in $L$. major R is mostly monosaccharide, disaccharide or trisaccharide (made up of $\beta$-d-Gal $p$ and $\beta$-D-Ara $p$ residues). ${ }^{2}$ The LPG changes in structure, principally in the doubling of phospho-
saccharide repeats, when the procyclic promastigote forms differentiate into metacyclic promastigote forms. ${ }^{3,4}$ LPG has been shown to be required for procyclic promastigote adhesion to the insect gut epithelium ${ }^{5,6}$ and for the successful invasion of macrophages. ${ }^{7}$ Biological, biochemical and biophysical experiments designed to probe the function, immunology, biosynthesis and conformation of LPG require the chemical synthesis of sub-structures of the LPG molecule. We now report the synthesis of the di-, tetra-, hexa- and octa-saccharide fragments 1-4 of the phosphoglycan portion of $L$. donovani LPG. All the synthetic oligomers contain a dec-9-enyl aglycone moiety and are designed to be used for both biosynthetic studies and the preparation of artificial antigens.

$$
\begin{aligned}
& \beta \text {-D-Galp-(1 } \rightarrow 4 \text { )- } \alpha \text {-D-Manp-(1-[-PO3 }{ }_{3} \mathrm{H}-6 \text { )- } \beta \text {-D-Galp- } \\
& \text { ( } 1 \rightarrow 4 \text { )- } \alpha \text {-D-Man } p-(1-]_{n}-\mathrm{O}\left[\mathrm{CH}_{2}\right]_{8} \mathrm{CH}=\mathrm{CH}_{2} \\
& 1 n=0 ; \quad 2 n=1 ; \quad 3 n=2 \\
& \alpha \text {-D-Manp-( } 1 \rightarrow 2 \text { )- } \alpha \text {-D-Man } p-\left(1-\left[-\mathrm{PO}_{3} \mathrm{H}-6\right)-\beta-\mathrm{d}-\mathrm{Gal} p-\right. \\
& (1 \rightarrow 4)-\alpha-\text { D-Manp- }(1-]_{3}-\mathrm{O}\left[\mathrm{CH}_{2}\right]_{8} \mathrm{CH}=\mathrm{CH}_{2} \\
& 4
\end{aligned}
$$

## Results and discussion

A retrosynthetic analysis of the most complicated oligomer, octaglycosyl triphosphate 4, shows that it can be prepared from the glycobiosyl H-phosphonate synthons 5 and 6 (for the consecutive introduction of the mannobiosyl phosphate and galactosylmannosyl phosphate fragments) and the monohydroxylic disaccharide synthon 7 . The synthons 6 and 7 can also be used to synthesize the shorter oligomers 1-3. The general approach for the preparation of the phosphorylated oligosaccharides 2-4 is based on the use of the glycosyl hydrogenphosphonate method ${ }^{8}$ for construction of the phosphodiester linkages and for stepwise elongation of the oligomeric chain.

The key galactosylmannosyl H-phosphonate block 6, containing a temporary dimethoxytrityl protecting group at the $6^{\prime}$-position, was prepared using acetobromogalactose 8 and 1,2,3,6-tetra- $O$-benzoyl- $\alpha$-D-mannopyranose 9 as the starting materials (Scheme 1). Compound 9 was synthesized in $64 \%$ yield (in addition to a small proportion of 1,2,3,6-tetra-O-benzoyl- $\beta$-D-mannopyranose 10) by selective benzoylation of D mannose with benzoyl chloride ( 4 mol equiv.; $-40^{\circ} \mathrm{C}$ ) in pyridine. Base-deficient glycosylation ${ }^{9}$ of the acceptor 9 with the galactopyranosyl bromide $\mathbf{8}$ in the presence of silver triflate and 2,4,6-trimethylpyridine (2,4,6-collidine) gave, after 20-30 min , the $\beta$-linked disaccharide 11 ( $74 \%$ ), a small proportion



5



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Retrosynthetic scheme
$(4 \%)$ of the $\alpha$-linked isomer 13 , and recovered acceptor $9(9.5 \%)$. If the glycosylation reaction was prolonged for 24 h , the disaccharide 12 ( $19 \%$ ), presumably formed from the disaccharide 11, was isolated in addition to the disaccharides 11 (47\%) and $13(7 \%)$. Condensation of the tetrabenzoate 9 and acetobromogalactose 8 in the presence of $\mathrm{Hg}(\mathrm{CN})_{2}-\mathrm{HgBr}_{2}$ in acetonitrile ${ }^{10}$ produced the disaccharides 11 and 13 in 54 and $32 \%$ yield, respectively.
The disaccharide 11 was converted into the $6^{\prime}-O$-dimethoxytrityl derivative 15 ( $72 \%$ ) by O -deacetylation ${ }^{11}$ with HCl in MeOH , followed by treatment of the resulting tetraol 14 first with dimethoxytrityl chloride in pyridine and then with benzoyl chloride in pyridine. The disaccharide 15 was selectively 1-Odebenzoylated with dimethylamine in acetonitrile ${ }^{8,12}$ to give the $\alpha$-hydroxy derivative 16 ( $77 \%$ ), which on phosphitylation ${ }^{8.12}$ with tri-imidazolylphosphine (prepared from $\mathrm{PCl}_{3}$, imidazole and $\mathrm{Et}_{3} \mathrm{~N}$ ) and mild hydrolysis gave the H phosphonate synthon 6 in $92 \%$ yield. Signals characteristic of the H-phosphonate group $\left[\delta_{\mathrm{P}} 1.57 ; \delta_{\mathrm{H}} 5.71\right.$ (dd, $J_{1.2} 1.8, J_{1 . \mathrm{P}}$ $8.85 \mathrm{~Hz}, 1-\mathrm{H}), 7.17\left(\mathrm{~d},{ }^{1} J_{\mathrm{H}, \mathrm{P}} 636 \mathrm{~Hz}, \mathrm{HP}\right.$ )] were present in the ${ }^{31} \mathrm{P}$ and ${ }^{1} \mathrm{H}$ NMR spectra of the disaccharide 6. The $\alpha-$
configuration followed from the characteristic value $(171 \mathrm{~Hz})$ of ${ }^{1} J_{\mathrm{C}, \mathrm{H}}$ for the signal of $\mathrm{C}-1\left(\delta_{\mathrm{C}} 92.61\right)$.
The monohydroxylic dec-9-enyl bioside 7 was prepared from acetobromogalactose 8 and dec- 9 -enyl 2,3,6-tri- $O$-benzoyl- $\alpha-$ Dmannopyranoside 20, as shown in Scheme 2. The mannoside 20 resulted from the glycosylation of dec-9-en-1-ol with acetobromomannose 17 in the presence of $\mathrm{Hg}(\mathrm{CN})_{2}-\mathrm{HgBr}_{2}$ in acetonitrile ${ }^{10}(\longrightarrow 18)$, followed by O -deacetylation and selective benzoylation ${ }^{13}$ of dec-9-enyl $\alpha$-D-mannopyranoside 19. Base-deficient galactosylation of the acceptor 20 with acetobromogalactose 8 (as above) gave the $\beta$-linked disaccharide $21(67 \%)$ and some of the $\alpha$-linked isomer $22(15 \%)$. The bioside 7 was prepared from the disaccharide 21 in an overall yield of $71 \%$ by O -deacetylation ${ }^{11}$ with $\mathrm{HCl}-\mathrm{MeOH}(\longrightarrow \mathbf{2 3}$ ), dimethoxytritylation of the 6 '-position and benzoylation ( $\longrightarrow \mathbf{2 4}$ ), followed by detritylation ( $\longrightarrow 7$ ) under mildly acidic conditions. Compound 21 also served as the direct precursor of the disaccharide 1 (see below).
To prepare the mannobiosyl H-phosphonate synthon 5 (Scheme 3), 1,3,4,6-tetra-O-acetyl- $\beta$-d-mannopyranose 26 was first glycosylated with benzobromomannose 25 under base-

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## Scheme 1

deficient conditions (as above) to give quantitatively the disaccharide 27, which was transformed into the H-phosphonate $5(92 \%)$ by selective 1-O-deacetylation with $\mathrm{Me}_{2} \mathrm{NH}$ in MeCN $(\longrightarrow 28)$, followed by H-phosphonylation. The structure of compound 5 was confirmed by fast-atom bombardment mass spectrometric $[\mathrm{FAB}-\mathrm{MS}(+)]\left(\mathrm{m} / \boldsymbol{z} \quad 1050.84,[\mathrm{M}+\mathrm{H}]^{+}\right)$and the ${ }^{31} \mathrm{P}$ and ${ }^{1} \mathrm{H}$ NMR data $\left[\delta_{\mathrm{P}} 1.88 ; \delta_{\mathrm{H}} 5.87\left(J_{1,2} 1.5 \mathrm{~Hz}, J_{1 . \mathrm{P}}\right.\right.$ $8.85 \mathrm{~Hz}, 1-\mathrm{H}), 7.01\left({ }^{1} J_{\mathrm{H}, \mathrm{P}} 632.6 \mathrm{~Hz}, \mathrm{HP}\right)$ ]. The $\alpha$-configuration of the reducing D-mannose residues in disaccharides 28 and 5 was evident from the characteristic values $(171$ and 170 Hz , respectively) of ${ }^{1} J_{\mathrm{C}, \mathrm{H}}$ for the signals of $\mathrm{C}-1$ (see Experimental section).

The chain-elongation cycle for synthesis of the oligo(glycobiosyl phosphates) 2-4 involved the coupling of a glycobiosyl Hphosphonate derivative with a hydroxylic acceptor, followed by oxidation of the resulting $H$-phosphonic diester to the phosphoric diester prior to removal of the temporary dimethoxytrityl protecting group. Oxidation of the H-phosphonic diester to the phosphoric diester during each elongation cycle is essential, since the higher stability of glycosyl phosphoric diesters permits selective deprotection and chromatographic isolation of the products, whereas the same operations with the glycosyl H-phosphonic diesters led to significant degradation. ${ }^{14,15}$

Condensation of the H-phosphonate 6 with the disaccharide 7 (Scheme 4) in pyridine in the presence of adamantane-1carbonyl chloride, followed by in situ oxidation with iodine in aq. pyridine and subsequent dedimethoxytritylation with $1 \% \mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}$ (TFA) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1 \mathrm{~min} ; 0^{\circ} \mathrm{C}\right)$, gave the tetrasaccharide phosphoric diester 29 in $81 \%$ overall yield. A similar sequence of reactions using either trimethylacetyl chloride or bis(2-oxooxazolidin-3-yl)phosphinic chloride ${ }^{16}$ as the condensing reagent resulted in the compound 29 in 79 and $70 \%$ yield, respectively. The hexasaccharide diphosphate derivative 30 was prepared in $75 \%$ yield from the $H-$ phosphonate 6 and the tetrasaccharide monophosphate 29 by using the prescribed route involving condensation (with adamantane-1-carbonyl chloride), oxidation and detritylation. In similar fashion, the octasaccharide triphosphate derivative 31 was obtained in $89 \%$ yield following the coupling between the mannobiosyl H-phosphonate 5 and the hexasaccharide derivative $\mathbf{3 0}$ and in situ oxidation.
The deprotected disaccharide 1 and oligo(glycobiosyl phosphates) 2-4 were prepared from the derivatives 21, 29, 30 and 31, respectively, by O-deacylation with $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ methanolic sodium methoxide.

The structures of the compounds 1-4 and 29-31 were confirmed by NMR and mass spectrometry data. The ${ }^{31} \mathrm{P}$ NMR data (see Experimental section) are characteristic of glycoside-linked phosphoric diesters. ${ }^{8,12,14,15}$ For the deprotected mono-, di- and tri-phosphates 2-4 in $\mathrm{D}_{2} \mathrm{O}$, the ${ }^{31} \mathrm{P}$ NMR spectra exhibited single signals at $\delta_{\mathbf{P}}-1.28,-1.29$ and -1.27 , respectively. However, the spectra of the protected di- and triphosphates 30 and 31 in $\mathrm{CDCl}_{3}$ consisted of two and three signals, respectively: $\delta_{\mathrm{P}}-2.99(\mathrm{P}),-3.08\left(\mathbf{P}^{\prime}\right)$ for diphosphate 30, and $-2.55\left(\mathbf{P}^{\prime \prime}\right),-2.95(\mathrm{P})$ and $32.02\left(\mathrm{P}^{\prime}\right)$ for triphosphate 31, indicating the non-equivalence of the phosphate groups in these oligomers (cf. ref. 8).
The presence of the ( $1 \rightarrow 6$ )-phosphodiester linkages was confirmed by the $\mathrm{C}-1$ and $\mathrm{C}-2$ signals of the corresponding $\mathrm{D}-$ mannose units and the $\mathrm{C}-5$ and $\mathrm{C}-6$ signals of the corresponding D-galactose units in the ${ }^{13} \mathrm{C}$ NMR spectra of compounds 2-4 (see Table 1). The signals were shifted as a result of the $\alpha$ - and $\beta$ effects of phosphorylation and were coupled with phosphorus. The $\alpha$-configuration of the D-mannosyl phosphate fragments followed from the positions of the C-3 and C-5 resonances of Man', Man" and Man". The chemical shifts of these signals were close to those of C-3 and C-5 of $\alpha$-D-mannopyranosyl phosphate, ${ }^{17}$ taking into account the influence of the glycosyl substituents at position-4 (of Man' and Man") and position-2 (of Man").

The molecular masses of the oligomers 1-4 and 29-31 were confirmed by electrospray $[\mathrm{ES}(-)]$ and $\mathrm{FAB}(+)$ mass spectrometry. The main signals in the spectra corresponded to the pseudo-molecular ions for the disaccharide $1(m / z 479.4$, $\left.[\mathrm{M}-\mathrm{H}]^{-}\right)$, the monophosphates $2\left(\mathrm{~m} / \mathrm{z} 883.4,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\right.\right.$ $\mathrm{H}]^{-}$) and $29\left(\mathrm{~m} / \mathrm{z} 2236.22,[\mathrm{M}+\mathrm{H}]^{+}\right.$), the diphosphates 3 ( $\mathrm{m} / \mathrm{z} 643.3,\left[\mathrm{M}-2 \mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}\right]^{2-}$ ) and $30(\mathrm{~m} / \mathrm{z}$ 1580.1, $[\mathrm{M}-$ $\left.2 \mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}\right]^{2-}$ ), and the triphosphates $4(m / z 563.3,[\mathrm{M}-3$ $\left.\mathrm{Et}_{3} \mathrm{~N}-3 \mathrm{H}\right]^{3-}$ ) and $31\left(m / z 1368.5\right.$, $\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-3 \mathrm{H}\right]^{3-}$ ).
To summarise, the first chemical syntheses of fragments (up to octasaccharide) of a natural antigenic phosphoglycan consisting of glycobiosyl phosphate units have been achieved using the glycosyl hydrogenphosphonate method.

## Experimental

## General procedures

Mps were determined on a Reichert hot-plate apparatus and are uncorrected. Optical rotations were measured with a PerkinElmer 141 polarimeter; $[\alpha]_{\mathrm{D}}$-values are given in units of $10^{-1}$ deg $\mathrm{cm}^{2} \mathrm{~g}^{-1}$. NMR spectra ( ${ }^{1} \mathrm{H}$ at 200 and 500 MHz , ${ }^{13} \mathrm{C}$ at 50.3




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Scheme 2

26
25

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and 125 MHz , and ${ }^{31} \mathrm{P}$ at 81 and 202.5 MHz ) were recorded with Bruker AM-200 and AM-500 spectrometers for solutions in $\mathrm{CDCl}_{3}$, unless otherwise indicated. Chemical shifts ( $\delta$ in ppm) are given relative to those for $\mathrm{Me}_{4} \mathrm{Si}$ (for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ ) and external aq. $85 \% \mathrm{H}_{3} \mathrm{PO}_{4}$ (for ${ }^{31} \mathrm{P}$ ); $J$ values are given in Hz . FAB mass spectra were recorded with a VG 70-250 SE mass spectrometer using an Ion-tech xenon gun. ES mass spectra were recorded with a VG Quattro system (VG Biotech, UK). TLC was performed on Polygram Sil G/UV ${ }_{254}$ (Macherey-Nagel, Germany) with $A$, toluene-ethyl acetate ( $9: 1$ ); $B$, toluene-ethyl acetate (8:2); $C$, toluene-ethyl acetate (7:3); $D$, ethyl acetatemethanol (9:1); $\quad E$, chloroform-methanol (95:5); $F$, chloroform-methanol (9:1); $G$, chloroform-methanol-water (93:7:0.35); $H$, chloroform-methanol-water ( $10: 10: 3$ ); and $I$, propan-2-ol-water ( $85: 15$ ) as developers and detection by charring with sulfuric acid-water-ethanol (15:85:5). Column chromatography was performed on Kieselgel 60 (0.040-0.063 mm ) (Merck). 1,3,4,6-Tetra- $O$-acetyl- $\beta$-D-mannopyranose, dec-9-en-1-ol, silver triflate, $p, p^{\prime}$-dimethoxytriphenylmethylchloride, adamantane-1-carbonyl chloride, and bis(2-oxooxazolidin-3yl)phosphinic chloride were purchased from Aldrich. Solutions worked up were concentrated under reduced pressure at $<40^{\circ} \mathrm{C}$.



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1,2,3,6-Tetra-O-benzoyl- $\alpha$-D-mannopyranose 9
Benzoyl chloride ( $9.27 \mathrm{~cm}^{3}, 80 \mathrm{mmol}$ ) was added dropwise to a stirred, cooled ( $-40^{\circ} \mathrm{C}$ ) solution of D-mannose ( $3.6 \mathrm{~g}, 20 \mathrm{mmol}$ ) in pyridine ( $40 \mathrm{~cm}^{3}$ ) during $c a .1 \mathrm{~h}$. The temperature was increased to $20^{\circ} \mathrm{C}$ for $3-4 \mathrm{~h}$, and the mixture then was stirred overnight. Most of the pyridine was evaporated off under reduced pressure, and a solution of the residue in chloroform was washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. Crystallisation of the residue from ethanol, and recrystallisation subsequently from ethyl acetate-hexane, gave the $\alpha$-tetrabenzoate 9 ( 6.14 g ). Column chromatography (solvent $A$ ) of the mother liquor gave an additional quantity of compound $9(1.5 \mathrm{~g}$; total yield 7.64 g , $64 \%$ ) $\mathrm{mp} 183-184^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{22}+41\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.64$ (solvent B) (Found: C, 68.6; H, 4.5. $\mathrm{C}_{34} \mathrm{H}_{28} \mathrm{O}_{10}$ requires C, $68.45 ; \mathrm{H}$, $4.7 \%) ; \delta_{\mathrm{H}} 3.20\left(1 \mathrm{H}, \mathrm{d}, J_{\text {он, } 4} 4.3, \mathrm{OH}\right), 4.25\left(1 \mathrm{H}, \mathrm{ddd}, J_{5,6 \mathrm{a}} 1.7,5-\right.$ $\mathrm{H}), 4.39\left(1 \mathrm{H}, \mathrm{dt}, J_{3.4}=J_{4,5}=10.0,4-\mathrm{H}\right), 4.55\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}}\right.$ $\left.12.5,6-\mathrm{H}^{\mathrm{a}}\right), 4.97\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 3.4,6-\mathrm{H}^{\mathrm{b}}\right), 5.80(2 \mathrm{H}, \mathrm{m}, 2$ - and 3H), $6.56\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.7,1-\mathrm{H}\right)$ and $7.25-8.16(20 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph})$. Continued elution gave 1,2,3,6-tetra-O-benzoyl- $\beta$-D-mannopyranose $10(0.8 \mathrm{~g}, 7 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}-42.5$ ( $c 1, \mathrm{CHCl}_{3}$ ); $R_{\mathrm{f}} 0.53$ (solvent $B$ ) (Found: C, $68.2 ; \mathrm{H}, 4.8 \%$ ); $\delta_{\mathrm{H}}$ $3.18\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{OH} .4} 4.5, \mathrm{OH}\right), 4.00\left(1 \mathrm{H}, \mathrm{ddd}, J_{\mathrm{S} .6 \mathrm{a}} 2.1,5-\mathrm{H}\right), 4.32$ $\left(1 \mathrm{H}, \mathrm{dt}, J_{3.4}=J_{4.5}=9.8,4-\mathrm{H}\right), 4.67\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.1,6-\mathrm{H}^{\mathrm{a}}\right)$, $4.96\left(1 \mathrm{H}, \mathrm{dd}, J_{5.6 \mathrm{~b}} 3.6,6-\mathrm{H}^{\mathrm{b}}\right), 5.52(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 5.99(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{2,3} 3.3,2-\mathrm{H}\right), 6.31\left(1 \mathrm{H}, \mathrm{d}, J_{1.2} 1.0,1-\mathrm{H}\right)$ and $7.30-8.25(20 \mathrm{H}, \mathrm{m}$, $4 \times \mathrm{Ph}$ ). The benzoylation reaction at $0^{\circ} \mathrm{C}$ (instead of at $-40^{\circ} \mathrm{C}$ ) resulted in the compounds 9 and 10 in 27 and $26 \%$ yield, respectively.

## 1,2,3,6-Tetra- $O$-benzoyl-4-O-(2,3,4,6-tetra-O-acetyl- $\beta$-D-galactopyranosyl)- $\alpha$-D-mannopyranose 11

Silver triflate ( $1.566 \mathrm{~g}, 6.092 \mathrm{mmol}$ ) was dried by evaporation of anhydrous toluene ( $2 \times 20 \mathrm{~cm}^{3}$ ) therefrom. A solution of acetobromogalactose $\mathbf{8}(2.17 \mathrm{~g}, 5.28 \mathrm{mmol})$, the tetrabenzoate 9 $(2.1 \mathrm{~g}, 3.52 \mathrm{mmol})$ and $2,4,6$-collidine ( $0.606 \mathrm{~cm}^{3}, 4.576 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(30 \mathrm{~cm}^{3}\right)$ was added dropwise to a stirred suspension of silver triflate in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(10 \mathrm{~cm}^{3}\right)$ at $-20^{\circ} \mathrm{C}$. Cooling was discontinued after the addition was complete, and, after 20 min , the mixture became slightly acidic; TLC (solvent $C$ ) then showed the formation of one major product ( $R_{\mathrm{f}} 0.42$ ). The mixture was neutralised with pyridine ( $1 \mathrm{~cm}^{3}$ ), filtered through a Celite pad, and the filtrate was concentrated. Column chromatography [toluene-ethyl acetate, $(95: 5) \rightarrow(80: 20)$ ] of the residue gave the disaccharide derivative $11(2.4 \mathrm{~g}, 74 \%)$; mp $107-109^{\circ} \mathrm{C}$ (from ethanol); $[\alpha]_{\mathrm{D}}^{23}+26\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.42$ (solvent $C$ ) (Found: $\mathrm{C}, 61.9 ; \mathrm{H}, 4.85 . \mathrm{C}_{48} \mathrm{H}_{46} \mathrm{O}_{19}$ requires C , $62.2 ; \mathrm{H}, 5.0 \%)$; $\delta_{\mathrm{H}} 1.78$ and $1.92(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{Ac}), 2.02(6 \mathrm{H}$, $\mathrm{s}, 2 \times \mathrm{Ac}), 3.51\left(2 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}\right.$ and $\left.6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 3.76\left(1 \mathrm{H}, \mathrm{dd}, J_{5} \cdot 6 \mathrm{~b} \cdot\right.$ $\left.5.8, J_{6 \mathrm{a}^{\prime} ; 6 \mathrm{~b}^{\prime}} 9.5,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 4.34\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5,6 \mathrm{a}} 3.0,5-\mathrm{H}\right), 4.52(1 \mathrm{H}$, $\left.\mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.0,6-\mathrm{H}^{\mathrm{a}}\right), 4.62\left(1 \mathrm{H}, \mathrm{t}, J_{3.4}=J_{4.5}=9.5,4-\mathrm{H}\right), 4.74$ ( $1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2} .7 .8,1^{\prime}-\mathrm{H}$ ), $4.80\left(1 \mathrm{H}, \mathrm{dd}, J_{5.6 \mathrm{~b}} 1.9,6-\mathrm{H}^{\mathrm{b}}\right), 4.90(1$ $\mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.5,3^{\prime}-\mathrm{H}$ ), 5.17 ( $1 \mathrm{H}, \mathrm{dd}, J_{2} \cdot 3^{\prime} \cdot 10.5,2^{\prime}-\mathrm{H}$ ), 5.18 ( 1 $\left.\mathrm{H}, \mathrm{br} \mathrm{d}, 4^{\prime}-\mathrm{H}\right), 5.86\left(1 \mathrm{H}, \mathrm{dd}, J_{2.3} 3.5,2-\mathrm{H}\right), 5.96(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H})$, $6.54\left(1 \mathrm{H}, \mathrm{d}, J_{1.2} 1.9,1-\mathrm{H}\right)$ and $7.33-8.15(20 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph})$; $\delta_{\mathrm{C}} 20.36,20.58$ and 20.73 ( MeCO ), 60.44 (C-6'), 62.29 (C-6), 66.53 (C-4'), 69.41 (C-2'), 69.42 (C-2), 70.39 (C-3), 70.75 (C-5'), 71.05 (C-3'), 71.68 (C-5), 73.51 (C-4), 91.40 (C-1), 101.25 (C-1'), 128.60-130.25 and 133.47-134.15 (Ph), 164.07, 165.03

Table $1 \quad{ }^{13} \mathrm{C}$ NMR data $\left(\mathrm{D}_{2} \mathrm{O}, \delta_{\mathrm{C}}\right.$ in ppm, $J$ in Hz ) for oligosaccharides $1-4\left(J_{\mathrm{C}, \mathrm{P}}\right.$-values in parentheses)

| Residue | Atom | 1 | $2^{\text {a }}$ | $3^{\text {a }}$ | $4^{a}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Man | C-1 | 100.94 | 100.76 | 100.54 | 100.51 |
|  | C-2 | 71.22 | 70.84 | 70.65 | 70.67 |
|  | C-3 | 70.89 | 70.84 | 70.65 | 70.67 |
|  | C-4 | 77.45 | 78.14 | 78.64 | 78.66 |
|  | C-5 | 72.46 | 72.34 | 72.33 | 72.33 |
|  | C-6 | 61.38 | 61.37 | 61.45 | 61.45 |
| Gal | C-1 | 104.33 | 104.42 | 104.46 | 104.49 |
|  | C-2 | 72.10 | 72.13 | 72.00 | 71.97 |
|  | C-3 | 74.02 | 73.69 | 73.64 | 73.51 |
|  | C-4 | 69.85 | 69.28 | 69.30 | 69.25 |
|  | C-5 | 76.51 | 74.89d (7.3) | 74.93d (7.5) | 74.84d (8.1) |
|  | C-6 | 62.32 | 65.53d (4.0) | 65.64d (5.5) | 65.43d (5.2) |
| Man' | C-1 |  | 97.12 d (5.4) | 97.03 d (5.5) | 96.99d (4.6) |
|  | C-2 |  | 71.08 d (8.3) | 71.07d (7.4) | 71.05d (7.4) |
|  | C-3 |  | 69.83 | 69.84 | 69.94 |
|  | C-4 |  | 77.05 | 78.17 | 78.22 |
|  | C-5 |  | 73.69 | 73.64 | 73.51 |
|  | C-6 |  | 61.37 | 61.45 | 61.45 |
| $\mathrm{Gal}^{\prime}$ | C-1 |  | 104.20 | 104.46 | 104.49 |
|  | C-2 |  | 72.13 | 72.00 | 71.97 |
|  | C-3 |  | 73.69 | 73.64 | 73.51 |
|  | C-4 |  | 69.83 | 69.30 | 69.25 |
|  | C-5 |  | 76.52 | 74.93d (7.5) | 74.84d (8.1) |
|  | C-6 |  | 62.30 | 65.64d (5.5) | 65.43 d (5.2) |
| Man" | C-1 |  |  | 97.03d (5.5) | 96.99 d (4.6) |
|  | C-2 |  |  | 71.07 d (7.4) | 71.05 d (7.4) |
|  | C-3 |  |  | 69.84 | 69.94 |
|  | C-4 |  |  | 77.04 | 78.22 |
|  | C-5 |  |  | 73.64 | 73.51 |
|  | C-6 |  |  | 61.45 | 61.45 |
| Gal" | C-1 |  |  | 104.20 | 104.49 |
|  | C-2 |  |  | 72.00 | 71.97 |
|  | C-3 |  |  | 73.64 | 73.51 |
|  | C-4 |  |  | 69.84 | 69.25 |
|  | C-5 |  |  | 76.52 | 74.84d (8.1) |
|  | C-6 |  |  | 62.27 | 65.43d (5.2) |
| Man' | C-1 |  |  |  | 95.85 d (4.6) |
|  | C-2 |  |  |  | 80.15d (7.4) |
|  | C-3 |  |  |  | 70.67 |
|  | C-4 |  |  |  | 67.83 |
|  | C-5 |  |  |  | 75.03 |
|  | C-6 |  |  |  | 62.03 |
| Man" ${ }^{\prime \prime}$ | C-1 |  |  |  | 103.35 |
|  | C-2 |  |  |  | 71.13 |
|  | C-3 |  |  |  | 71.47 |
|  | C-4 |  |  |  | 67.83 |
|  | C-5 |  |  |  | 74.40 |
|  | C-6 |  |  |  | 62.03 |
| Dec-9-en-1-yl | $=\mathrm{CH}_{2}$ | 115.29 | 115.29 | 115.17 | 115.13 |
|  | $-\mathrm{CH}=$ | 140.11 | 140.56 | 140.64 | 141.46 |
|  | $\mathrm{OCH}_{2}$ | $68.91$ | $69.11$ | $69.30$ | $69.25$ |
|  | $\mathrm{CCH}_{2} \mathrm{C}$ | 27.24, 29.91-30.68, 34.89 | 26.90, 29.74-30.21, 34.65 | 26.51, 29.20-29.62, 34.27 | 26.51, 29.29-29.63, 34.30 |

${ }^{a}$ Additional signals of $\mathrm{Et}_{3} \mathrm{NH}^{+}\left(\delta_{\mathrm{C}} 9.40-9.45\right.$ and $\left.\delta_{\mathrm{C}} 47.82-47.85\right)$ were present.
and $165.92\left(\mathrm{PhCO}_{2}\right), 169.43,169.97$ and $170.15\left(\mathrm{MeCO}_{2}\right)$. Also isolated were the acceptor $9(0.2 \mathrm{~g}, 9.5 \%$ recovery ) and 1,2,3,6-tetra-O-benzoyl-4-O-(2,3,4,6-tetra-O-acetyl- $\alpha$-D-galacto-pyranosyl)- $\alpha$-D-mannopyranose $13(0.13 \mathrm{~g}, 4 \%)$; amorphous solid; $[\alpha]_{\mathrm{D}}^{23}+74\left(c 1.23, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.49$ (solvent $C$ ) (Found: $\mathrm{C}, 61.9 ; \mathrm{H}, 4.7 \%) ; \delta_{\mathrm{H}} 1.80,1.94,1.95$ and $2.07(12 \mathrm{H}, 4 \times \mathrm{s}$, $4 \times \mathrm{Ac}), 3.90\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{a}^{\prime}} 5.8,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.04\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}}\right.$ $\left.11.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 4.36\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{~b}^{\prime}} 7.4,5^{\prime}-\mathrm{H}\right), 4.44\left(1 \mathrm{H}\right.$, ddd, $J_{5,6 \mathrm{a}}$ $3.6,5-\mathrm{H}), 4.61\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 13.0,6-\mathrm{H}^{\mathrm{a}}\right), 4.68\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=\right.$ $\left.J_{4,5}=9.0,4-\mathrm{H}\right), 4.86\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 2.0,6-\mathrm{H}^{\mathrm{b}}\right), 5.21(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{2^{\prime}, 3^{\prime}} 11.0,2^{\prime}-\mathrm{H}\right), 5.34\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.0,3^{\prime}-\mathrm{H}\right), 5.45(1 \mathrm{H}, \mathrm{br} \mathrm{d}$, $\left.4^{\prime}-\mathrm{H}\right), 5.59\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 3.8,1^{\prime}-\mathrm{H}\right), 5.86\left(2 \mathrm{H}, \mathrm{m}, J_{2,3} 3.0,2\right.$ - and $3-\mathrm{H}), 6.57\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.5,1-\mathrm{H}\right)$ and $7.32-8.21(20 \mathrm{H}, \mathrm{m}$, $4 \times \mathrm{Ph}) ; \delta_{\mathrm{C}} 20.40$ and $20.63(\mathrm{MeCO}), 61.21\left(\mathrm{C}-6^{\prime}\right), 63.23(\mathrm{C}-6)$, 66.82 (C-2'), 67.37 (C-3'), 67.71 (C-5'), 67.76 (C-4'), 69.20 (C-2), 71.63 (2 C, C-4 + -5), 72.70 (C-3), 91.46 (C-1), 97.66
(C-1'), 128.31-130.24 and 133.45-134.11 (Ph), 164.18, 165.03, 165.50 and $165.98\left(\mathrm{PhCO}_{2}\right), 169.63,169.86,170.08$ and 170.30 $\left(\mathrm{MeCO}_{2}\right) ;$ ES-MS(+) data: $m / z 949.3\left(100 \%,[\mathrm{M}+\mathrm{Na}]^{+}\right)$ $\left(\mathrm{C}_{48} \mathrm{H}_{46} \mathrm{O}_{19}\right.$ requires $\mathrm{M}, 926.26$ ).

## 1,2,3,6-Tetra-O-benzoyl-4-O-[2,3,4-tri- $O$-benzoyl-6- $O$-( $p, p^{\prime}$ -

 dimethoxytrityl)- $\beta$-D-galactopyranosyl]- $\alpha$-D-mannopyranose 15 A solution of HCl in MeOH [prepared at $0^{\circ} \mathrm{C}$ from acetyl chloride ( $0.4 \mathrm{~cm}^{3}$ ) and methanol $\left.\left(10 \mathrm{~cm}^{3}\right)\right]$ was added to a solution of compound $11(1.41 \mathrm{~g}, 1.52 \mathrm{mmol})$ in $\mathrm{CHCl}_{3}\left(3 \mathrm{~cm}^{3}\right)$, and the resulting solution was kept at $20^{\circ} \mathrm{C}$ for $40-43 \mathrm{~h}$; TLC (solvent $F$ ) then showed the formation of one major product ( $R_{\mathrm{f}}$ 0.48 ; presumably the O-deacetylated derivative 14). Ethanol ( 10 $\mathrm{cm}^{3}$ ) was added to the reaction mixture, which was neutralised with anhydrous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and filtered, and the solids were washed with ethanol. The filtrate and washings wereconcentrated, and pyridine ( $3 \times 15 \mathrm{~cm}^{3}$ ) was evaporated from the residue. The residue was dissolved in pyridine ( $15 \mathrm{~cm}^{3}$ ), p, p'dimethoxytriphenylmethyl chloride $(0.7 \mathrm{~g}, 2.07 \mathrm{mmol})$ was added, and the solution was kept for 48 h at $20^{\circ} \mathrm{C}$ before benzoyl chloride ( $1 \mathrm{~cm}^{3}, 8.6 \mathrm{mmol}$ ) was also added to the stirred mixture at $-10^{\circ} \mathrm{C}$. After 16 h at $20^{\circ} \mathrm{C}$, the reaction mixture was diluted with $\mathrm{CHCl}_{3}$ and washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. Column chromatography [hexane-ethyl acetate, (3:2)] gave the dimethoxytrityl derivative $15(1.5 \mathrm{~g}, 72 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+62.7\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.5$ (solvent $A$ ), 0.74 (solvent $B$ ) (Found: C, 72.0; H, 5.2. $\mathrm{C}_{82} \mathrm{H}_{68} \mathrm{O}_{20}$ requires $\mathrm{C}, 71.7 ; \mathrm{H}$, $5.0 \%) ; \delta_{\mathrm{H}} 3.13\left(1 \mathrm{H}, \mathrm{t}, J_{5 \cdot 6 \mathrm{a}^{\prime}}=J_{6 \mathrm{a}^{\prime} \cdot 6 \mathrm{~b}^{\prime}}=8.6,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 3.28(1 \mathrm{H}$, dd, $J_{5^{\prime} .6 \mathrm{~b}^{\prime}} 5.4,6^{\prime}-\mathrm{H}^{\mathrm{b}}$ ), 3.63 and $3.64(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{MeO}$ ), 3.86 $(1 \mathrm{H}, \mathrm{br}$ dd, $5 '-\mathrm{H}), 4.20(1 \mathrm{H}, \mathrm{br} \mathrm{d}, 5-\mathrm{H}), 4.59\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right)$, $4.68\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.5,4-\mathrm{H}\right), 5.0\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 7.8,1^{\prime}-\right.$ H), $5.52\left(1 \mathrm{H}, \mathrm{dd}, J_{3 ; 4} \cdot 3.0,3^{\prime}-\mathrm{H}\right), 5.69\left(1 \mathrm{H}, \mathrm{dd}, J_{2 \cdot 3} \cdot 10.4,2^{\prime}-\right.$ H), 5.85 ( $1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.2,2-\mathrm{H}$ ), $5.91(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 6.04$ ( 1 H , dd, $\left.J_{4^{\prime} .5^{\prime}} \sim 0.5,4^{\prime}-\mathrm{H}\right), 6.49\left(1 \mathrm{H}, \mathrm{d}, J_{1.2} 1.9,1-\mathrm{H}\right)$ and $6.59(2 \mathrm{H}$, d) and 6.90-8.15 ( $46 \mathrm{H}, \mathrm{m}$ ) $\left(2 \times \mathrm{C}_{6} \mathrm{H}_{4}, 8 \times \mathrm{Ph}\right) ; \delta_{\mathrm{C}} 55.18$ ( MeO ), $59.64\left(\mathrm{C}-6^{\prime}\right), 62.22(\mathrm{C}-6), 67.66$ (C-4'), 69.71 (C-2), 70.28 (C-3), 70.38 (C-2'), 72.10 (C-5), 72.26 (C-3'), 72.71 (C-5'), 72.80 (C-4), $86.40\left(\mathrm{Ar}_{3} \mathrm{C}\right), 91.34(\mathrm{C}-1), 101.55\left(\mathrm{C}-1^{\prime}\right), 113.14$, $125.44-130.27,133.21-135.77,144.12,158.42$ and 158.52 $\left(\mathrm{C}_{6} \mathrm{H}_{4}\right.$ and Ph$)$ and 164.12-165.82 ( $\mathrm{PhCO}_{2}$ ).

## 2,3,6-Tri-O-benzoyl-4-O-[2,3,4-tri-O-benzoyl-6-O- (p,p'-

 dimethoxytrityl)- $\beta$-D-galactopyranosyl]- $\alpha$-D-mannopyranose 16 Anhydrous dimethylamine ( $0.2 \mathrm{~cm}^{3}, 3.0 \mathrm{mmol}$ ) was added to a solution of compound $15(0.686 \mathrm{~g}, 0.5 \mathrm{mmol})$ in $\mathrm{MeCN}\left(5 \mathrm{~cm}^{3}\right)$ at $-20^{\circ} \mathrm{C}$, and the mixture was kept at $20^{\circ} \mathrm{C}$ with monitoring by TLC (solvent $B$ ). After 27 h (note: compound 15 was not allowed to be consumed completely), the mixture was concentrated to dryness, and acetonitrile was evaporated off from the residue. Column chromatography [toluene-ethyl acetate, $(95: 5) \rightarrow(80: 20)]$ gave unchanged $15(0.081 \mathrm{~g}, 12 \%$ recovery) and 1-hydroxy derivative $16(0.489 \mathrm{~g}, 77 \%)$; amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+31.5\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.44$ (solvent $B$ ) (Found: C, 70.7; H, 5.15. $\mathrm{C}_{75} \mathrm{H}_{64} \mathrm{O}_{19}$ requires $\mathrm{C}, 71.0 ; \mathrm{H}$, $5.1 \%) ; \delta_{\mathrm{H}} 3.15\left(1 \mathrm{H}, \mathrm{t}, J_{5^{\prime}, 6 \mathrm{a}^{\prime}}=J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}}=8.8,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 3.28(1 \mathrm{H}$, dd, $J_{5^{\prime}, 66^{\prime}} 4.6,6^{\prime}-\mathrm{H}^{\mathrm{b}}$ ), 3.64 and $3.65(6 \mathrm{H}, 2 \times \mathrm{s}, 2 \times \mathrm{MeO})$, $3.76\left(1 \mathrm{H}, \mathrm{br}\right.$ dd, $\left.5^{\prime}-\mathrm{H}\right), 3.89\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{OH}, 1} 3.6, \mathrm{OH}\right), 4.29(1 \mathrm{H}, \mathrm{br}$ $\mathrm{d}, 5-\mathrm{H}), 4.52\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right), 4.61\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.7,4-\right.$ H), $4.95\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2} .7 .7,1^{\prime}-\mathrm{H}\right), 5.32\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 1.5,1-\mathrm{H}\right)$, $5.49\left(1 \mathrm{H}, \mathrm{dd}, J_{3.4} \cdot 3.0,3^{\prime}-\mathrm{H}\right), 5.64(1 \mathrm{H}, \mathrm{dd}, 2-\mathrm{H}), 5.65(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{2 \cdot \cdot 3} \cdot 10.6,2^{\prime}-\mathrm{H}\right), 5.84\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 2.7,3-\mathrm{H}\right), 5.98\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, 4^{\prime}-\right.$ $\mathrm{H})$ and $6.55\left(2 \mathrm{H}\right.$, d) and $7.10-8.01(41 \mathrm{H}, \mathrm{m})\left(2 \times \mathrm{C}_{6} \mathrm{H}_{4}\right.$, $7 \times \mathrm{Ph}) ; \delta_{\mathrm{C}} 55.0(\mathrm{MeO}), 59.65\left(\mathrm{C}-6^{\prime}\right), 62.56(\mathrm{C}-6), 67.57\left(\mathrm{C}-4^{\prime}\right)$, 69.49 (C-5), 69.68 (C-3), 70.26 (C-2'), 71.04 (C-2), 71.95 (C-3'), 72.44 (C-5'), $72.94(\mathrm{C}-4), 86.26$ ( $\mathrm{Ar}_{3} \mathrm{C}$ ), $91.99(\mathrm{C}-1), 100.88$ (C$1^{\prime}$ ), 112.99, 125.44-130.26, 133.23-135.77, 144.20, 158.41 and $158.51\left(\mathrm{C}_{6} \mathrm{H}_{4}\right.$ and Ph$)$ and $165.09-166.17\left(\mathrm{PhCO}_{2}\right)$.
## Triethylammonium 2,3,6-tri- $O$-benzoyl-4-O-[2,3,4-tri- $O$ -benzoyl-6- $O$-( $p, p^{\prime}$-dimethoxytrityl)- $\beta$-D-galactopyranosyl]- $\alpha$-Dmannopyranosyl hydrogenphosphonate 6

To a stirred solution of imidazole ( $0.456 \mathrm{~g}, 6.72 \mathrm{mmol}$ ) in $\mathrm{MeCN}\left(12 \mathrm{~cm}^{3}\right)$ at $0{ }^{\circ} \mathrm{C}$ was added phosphorus trichloride $\left(0.178 \mathrm{~cm}^{3}, 2.02 \mathrm{mmol}\right)$ and then triethylamine $\left(0.98 \mathrm{~cm}^{3}, 7.06\right.$ mmol ). The mixture was stirred for 15 min , after which a solution of compound $16(0.585 \mathrm{~g}, 0.461 \mathrm{mmol})$ in MeCN ( 12 $\mathrm{cm}^{3}$ ) was added dropwise during 30 min at $0^{\circ} \mathrm{C}$. The mixture was stirred at $20^{\circ} \mathrm{C}$ for $10-15 \mathrm{~min}$ and quenched with 1 mol $\mathrm{dm}^{-3}$ triethylammonium (TEA) hydrogen carbonate ( $\mathrm{pH} 7 ; 2.8$ $\mathrm{cm}^{3}$ ). The clear solution was stirred for $15 \mathrm{~min}, \mathrm{CHCl}_{3}(150$ $\mathrm{cm}^{3}$ ) was added, and the organic layer was washed in turn with ice-water ( $2 \times 80 \mathrm{~cm}^{3}$ ) and cold $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ TEA hydrogen carbonate ( $2 \times 80 \mathrm{~cm}^{3}$ ), dried by filtration through cotton
wool, and concentrated. Column chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\right.$ $\left.\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N},(98: 1: 1) \rightarrow(93: 6: 1)\right]$ gave the biosyl H-phosphonate $6(0.607 \mathrm{~g}, 92 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+24(c 1$, $\left.\mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.44$ (solvent $F$ ); $\delta_{\mathrm{H}} 1.30\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right), 3.00$ $\left(6 \mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 3.10\left(1 \mathrm{H}, \mathrm{t}, J_{5 \cdot 6 \mathrm{a}^{\prime}}=J_{6 \mathrm{a}^{\prime}: 6 \mathrm{~b}^{\prime}}=8.6,6^{\prime}-\right.$ $\left.\mathrm{H}^{\mathrm{a}}\right), 3.22\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{~b}^{\prime}} 5.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 3.64$ and $3.65(6 \mathrm{H}, 2 \mathrm{~s}$, $2 \times \mathrm{MeO}), 3.76\left(1 \mathrm{H}, \mathrm{br}\right.$ dd, $\left.5^{\prime}-\mathrm{H}\right), 4.30\left(1 \mathrm{H}, \mathrm{dt}, J_{5,6} 2.2,5-\mathrm{H}\right)$, $4.49\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}\right), 4.55\left(2 \mathrm{H}, \mathrm{d}, 6-\mathrm{H}_{2}\right), 4.89(1$ $\left.\mathrm{H}, \mathrm{d}, J_{1 ; 2}, 7.6,1^{\prime}-\mathrm{H}\right), 5.46\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.8,3^{\prime}-\mathrm{H}\right), 5.59(1 \mathrm{H}$, dd, $\left.J_{2^{\prime}, 3^{\prime}} 10.2,2^{\prime}-\mathrm{H}\right), 5.66\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.2,2-\mathrm{H}\right), 5.71(1 \mathrm{H}, \mathrm{dd}$, $J_{1,2} 1.8, J_{1, \mathrm{P}} 8.9,1-\mathrm{H}$ ), $5.76(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 5.97\left(1 \mathrm{H}, \mathrm{dd}, J_{4}{ }^{\prime} 5^{\prime}\right.$ $\left.\sim 0.5,4^{\prime}-\mathrm{H}\right), 7.17\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{H} . \mathrm{P}} 636.0, \mathrm{HP}\right)$ and $6.54(2 \mathrm{H}, \mathrm{d})$ and 6.90-8.00 ( $41 \mathrm{H}, \mathrm{m}$ ) $\left(2 \times \mathrm{C}_{6} \mathrm{H}_{4}, 7 \times \mathrm{Ph}\right)$; $\delta_{\mathrm{C}} 8.50$ and 45.58 ( Et ), $54.98(\mathrm{MeO}), 59.42\left(\mathrm{C}-6^{\prime}\right), 62.52(\mathrm{C}-6), 67.48$ (C-4'), 70.12 (C-3), 70.15 (C-5), 70.40 (C-2'), 70.87 (d, $J_{\mathrm{C}, \mathrm{P}} 6.9, \mathrm{C}-2$ ), 71.93 (C$\left.3^{\prime}\right), 72.35$ (C-5'), 72.58 (C-4), $86.17\left(\mathrm{Ar}_{3} \mathrm{C}\right), 92.61$ (d, $J_{\mathrm{C}, \mathrm{P}} 3.6$, $\left.{ }^{1} J_{\mathrm{C}, \mathrm{H}} 171, \mathrm{C}-1\right), 100.99\left(\mathrm{C}-1{ }^{1}\right), 112.95,126.65-130.06,132.68$ $135.62,143.95,158.22$ and $158.32\left(\mathrm{C}_{6} \mathrm{H}_{4}\right.$ and Ph$)$ and 164.79$165.42\left(\mathrm{PhCO}_{2}\right) ; \delta_{\mathrm{P}} 1.57$.

## Dec-9-enyl 2,3,4,6-tetra- $O$-acetyl- $\alpha$-D-mannopyranoside 18

A solution of acetobromomannose $17^{10}(4.11 \mathrm{~g}, 10 \mathrm{mmol})$ in $\mathrm{MeCN}\left(15 \mathrm{~cm}^{3}\right)$ was added dropwise to a stirred mixture of dec9 -en-1-ol ( $1.04 \mathrm{~g}, 6.67 \mathrm{mmol}), \mathrm{Hg}(\mathrm{CN})_{2}(2.53 \mathrm{~g}, 10.0 \mathrm{mmol})$ and $\mathrm{HgBr}_{2}(0.5 \mathrm{~g})$ in $\mathrm{MeCN}\left(30 \mathrm{~cm}^{3}\right)$. The mixture was stirred at $20^{\circ} \mathrm{C}$ overnight, then was concentrated under reduced pressure, and the residue was dissolved in $\mathrm{CHCl}_{3}$. The suspension was filtered to remove mercury salts, and the filtrate was washed successively with $1 \mathrm{~mol} \mathrm{dm}{ }^{-3} \mathrm{KBr}$ and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. Column chromatography [toluene-ethyl acetate, $(97: 3) \rightarrow(90: 10)]$ of the residue gave the decenyl glycoside $18(2.3 \mathrm{~g}, 71 \%)$ as a syrup; $[\alpha]_{\mathrm{D}}^{22}+40\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}}$ 0.67 (solvent $C$ ) (Found: C, 59.55; H, 8.0. $\mathrm{C}_{24} \mathrm{H}_{38} \mathrm{O}_{10}$ requires $\mathrm{C}, 59.25 ; \mathrm{H}, 7.9 \%) ; \delta_{\mathrm{H}} 1.32\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.60(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2}$ ), 1.97, 2.09 and $2.15(9 \mathrm{H}, 3 \times \mathrm{s}, 3 \times \mathrm{Ac}), 2.03(5 \mathrm{H}, \mathrm{m}, \mathrm{Ac}$ and $\left.\mathrm{CH}_{2}\right), 3.45$ and $3.67\left(2 \mathrm{H}, 2 \times \mathrm{dt},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 10.0,{ }^{3} J_{\mathrm{H}, \mathrm{H}} 7.0\right.$, $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.98\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5,6 \mathrm{a}} 2.5,5-\mathrm{H}\right), 4.11\left(1 \mathrm{H}\right.$, dd, $J_{6 \mathrm{a}, 6 \mathrm{~b}}$ $\left.12.5,6-\mathrm{H}^{\mathrm{a}}\right), 4.28\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 5.5,6-\mathrm{H}^{\mathrm{b}}\right.$ ), $4.60\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.5\right.$, $1-\mathrm{H}), 4.93\left(1 \mathrm{H}, \mathrm{dd},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.8,{ }^{3} J_{\mathrm{H}, \mathrm{H}} 10.0, \mathrm{CH}=\mathrm{CH}_{2}\right), 4.99(1 \mathrm{H}$, $\mathrm{dd},{ }^{3} J_{\mathrm{H}, \mathrm{H}} 17.5, \mathrm{CH}=\mathrm{CH}_{2}$ ), $5.23\left(1 \mathrm{H}\right.$, dd, $J_{2,3} 3.6,2-\mathrm{H}$ ), 5.27 ( 1 $\left.\mathrm{H}, \mathrm{t}, J_{3,4}=J_{4.5}=10.0,4-\mathrm{H}\right), 5.35(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H})$ and $5.81[1$ $\mathrm{H}, \mathrm{ddt}, \mathrm{J}\left(\mathrm{H}, \mathrm{CH}_{2}\right) 6.5, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ].

## Dec-9-enyl $\alpha$-D-mannopyranoside 19

A solution of the tetraacetate $18(3.26 \mathrm{~g}, 6.71 \mathrm{mmol})$ in MeOH ( $30 \mathrm{~cm}^{3}$ ) and tetrahydrofuran $\left(8 \mathrm{~cm}^{3}\right.$ ) was treated with 4.6 mol $\mathrm{dm}^{-3} \mathrm{NaOMe}$ in $\mathrm{MeOH}\left(1 \mathrm{~cm}^{3}\right)$ overnight at $20^{\circ} \mathrm{C}$, and was then deionised with Dowex 50W-X4 $\left(\mathrm{H}^{+}\right)$resin, and concentrated to give the decenyl mannoside $19(2.13 \mathrm{~g}, 99.7 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+56(c 0.5, \mathrm{MeOH}) ; R_{\mathrm{f}} 0.5$ (solvent D) (Found: C, $60.1 ; \mathrm{H}, 9.5 . \mathrm{C}_{16} \mathrm{H}_{30} \mathrm{O}_{6}$ requires $\mathrm{C}, 60.4 ; \mathrm{H}$, $9.5 \%) ; \delta_{\mathrm{C}}\left(\mathrm{CD}_{3} \mathrm{OD}+\mathrm{CDCl}_{3}\right) 26.30,29.05,29.11,29.50,29.62$ and $33.81\left(\mathrm{CH}_{2}\right), 61.75(\mathrm{C}-6), 67.45(\mathrm{C}-4), 67.72\left({ }^{1} J_{\mathrm{C}, \mathrm{H}} 140.0\right.$, $\mathrm{OCH}_{2} \mathrm{CH}_{2}$ ), $71.22(\mathrm{C}-2), 71.73(\mathrm{C}-3), 73.15(\mathrm{C}-5), 100.41\left({ }^{1} J_{\mathrm{C}, \mathrm{H}}\right.$ 168.5, $\mathrm{C}-1), 113.91\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$ and $139.09\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$.

## Dec-9-enyl 2,3,6-tri- $O$-benzoyl- $\alpha$-D-mannopyranoside 20

Benzoyl chloride ( $2.57 \mathrm{~cm}^{3}, 22.14 \mathrm{mmol}$ ) was added dropwise over a period of 30 min to a cooled $\left(-40^{\circ} \mathrm{C}\right)$ and stirred solution of the mannoside $19(2.13 \mathrm{~g}, 6.69 \mathrm{mmol})$ in pyridine ( 20 $\mathrm{cm}^{3}$ ). The temperature was increased to $20^{\circ} \mathrm{C}$ for 3 h , and stirring was continued overnight. Most of the pyridine was evaporated off under reduced pressure. A solution of the residue in $\mathrm{CHCl}_{3}$ was washed in turn with saturated aq. $\mathrm{NaHCO}_{3}$ and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. Column chromatography [toluene-ethyl acetate, $(100: 0) \rightarrow$ ( $95: 5$ )] of the residue gave the tribenzoate $20(2.16 \mathrm{~g}, 51 \%)$; $\mathrm{mp} 85-87^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{22}+12.5\left(c \mathrm{l}, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.62$ (solvent
A), 0.74 (solvent $B$ ) (Found: C, 70.2; $\mathrm{H}, 6.75 . \mathrm{C}_{37} \mathrm{H}_{42} \mathrm{O}_{9}$ requires $\mathrm{C}, 70.5 ; \mathrm{H}, 6.7 \%)$; $\delta_{\mathrm{H}} 1.30\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.66(2$ H , quintet, $J 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ), $2.05(2 \mathrm{H}$, quartet, $J 6.5$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=$ ), $3.15\left(1 \mathrm{H}, \mathrm{d}, J_{\mathrm{OH}, 4} 4.8, \mathrm{OH}\right.$ ), 3.52 and 3.78 ( 2 H , $\left.2 \times \mathrm{dt},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 9.5,{ }^{3} J_{\mathrm{H}, \mathrm{H}} 6.8, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 4.12\left(1 \mathrm{H}, \mathrm{ddd}, J_{5,6 \mathrm{a}}\right.$ $1.5,5-\mathrm{H}), 4.28\left(1 \mathrm{H}, \mathrm{dt}, J_{3,4}=J_{4.5}=9.5,4-\mathrm{H}\right), 4.65(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.8,6-\mathrm{H}^{\mathrm{a}}\right), 4.87\left(1 \mathrm{H}, \mathrm{dd}, J_{5.6 \mathrm{~b}} 3.9,6-\mathrm{H}^{\mathrm{b}}\right), 4.93(1 \mathrm{H}, \mathrm{dd}$, $\left.{ }^{2} J_{\mathrm{H}, \mathrm{H}} 1.5,{ }^{3} J_{\mathrm{H}, \mathrm{H}} 10.2, \mathrm{CH}=\mathrm{C} \mathrm{H}_{2}\right), 5.00\left(1 \mathrm{H}, \mathrm{dd},{ }^{3} J_{\mathrm{H}, \mathrm{H}} 16.5\right.$, $\mathrm{CH}=\mathrm{CH}_{2}$ ), $5.01\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.5,1-\mathrm{H}\right), 5.61\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3: 1,2-\right.$ $\mathrm{H}), 5.65(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 5.82\left[1 \mathrm{H}\right.$, ddt, $J\left(\mathrm{H}, \mathrm{CH}_{2}\right) 6.5$, $\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ] and $7.30-8.14$ ( $15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph}$ ); $\delta_{\mathrm{c}} 26.08$, 28.86, 29.01, 29.28, 29.34 and $33.72\left(\mathrm{CH}_{2}\right), 63.47(\mathrm{C}-6), 66.44$ (C-4), $68.45\left(\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 70.70(\mathrm{C}-2), 71.22(\mathrm{C}-5), 72.83(\mathrm{C}-3)$, 97.60 ( $\left.{ }^{1} J_{\mathrm{C}, \mathrm{H}} 171, \mathrm{C}-1\right)$, $114.09\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$, 128.29-129.87, $133.12-133.28(\mathrm{Ph}), 139.13\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 165.34,166.69$ and $166.81\left(\mathrm{PhCO}_{2}\right)$.

## Dec-9-enyl 2,3,6-tri- $O$-benzoyl-4- $O$-(2,3,4,6-tetra- $O$-acetyl- $\beta$ -D-galactopyranosyl)- $\alpha$-D-mannopyranoside 21

The reaction of compounds $8(0.822 \mathrm{~g}, 2.0 \mathrm{mmol})$ and $20(0.63 \mathrm{~g}$, 1.0 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(15 \mathrm{~cm}^{3}\right)$ in the presence of $\mathrm{AgO}_{3} \mathrm{SCF}_{3}$ ( $0.6 \mathrm{~g}, 2.33 \mathrm{mmol}$ ) and $2,4,6$-collidine ( $0.229 \mathrm{~cm}^{3}, 1.73 \mathrm{mmol}$ ) was accomplished at $-20 \rightarrow+20^{\circ} \mathrm{C}$, as described for the preparation of the disaccharide 11. Column chromatography (solvent $A$ ) gave the disaccharide derivative $21(0.638 \mathrm{~g}, 67 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}-10\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.37$ (solvent B) (Found: C, 63.8; H, 6.2. $\mathrm{C}_{51} \mathrm{H}_{60} \mathrm{O}_{18}$ requires $\mathrm{C}, 63.7 ; \mathrm{H}$, $6.3 \%) ; \delta_{\mathrm{H}} 1.34\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.65(2 \mathrm{H}$, quintet, $J 6.5$, $\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ), 1.79, 1.93, 2.00 and $2.01(12 \mathrm{H}, 4 \times \mathrm{s}$, $4 \times \mathrm{Ac}), 2.05\left(2 \mathrm{H}\right.$, quartet, $J 6.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=$ ), 3.35-3.57 ( 3 $\mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2}, 5^{\prime}-\mathrm{H}$ and $6^{\prime}-\mathrm{H}^{\mathrm{a}}$ ), $3.73\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{~b}^{\prime}} 6.9\right.$, $\left.J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 10.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 3.78\left(1 \mathrm{H}, \mathrm{dt},{ }^{2} J_{\text {H., }} 9.6,{ }^{3} J_{\text {H. H }} 6.7\right.$, $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 4.21\left(1 \mathrm{H}, \mathrm{ddd}, J_{5,6 \mathrm{a}} 4.3,5-\mathrm{H}\right), 4.45\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=\right.$ $\left.J_{4.5}=9.0,4-\mathrm{H}\right), 4.50\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a} .6 \mathrm{~b}} 12.0,6-\mathrm{H}^{\mathrm{a}}\right), 4.70(1 \mathrm{H}, \mathrm{d}$, $\left.J_{1^{\prime}, 2^{\prime}} 7.9,1^{\prime}-\mathrm{H}\right), 4.82\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 2.0,6-\mathrm{H}^{\mathrm{b}}\right), 4.85-5.05(4 \mathrm{H}$, $\mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}, 1-$ and $\left.3^{\prime}-\mathrm{H}\right), 5.15\left(1 \mathrm{H}\right.$, br d, $\left.J_{3^{\prime}, 4}, 2.5,4^{\prime}-\mathrm{H}\right), 5.16$ ( $\left.1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3}, 10.5,2^{\prime}-\mathrm{H}\right), 5.64\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 1.9,2-\mathrm{H}\right), 5.81(1$ H , dd, $\left.J_{2,3} 3.5,3-\mathrm{H}\right), 5.82\left[1 \mathrm{H}\right.$, ddt, $J_{\mathrm{H}, \mathrm{H}} 10.1, J_{\mathrm{H}, \mathrm{H}} 16.7$, $J\left(\mathrm{H}_{,} \mathrm{CH}_{2}\right) 6.5, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ] and $7.30-8.15(15 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{Ph})$; $\delta_{\mathrm{C}} 20.40,20.58$ and 20.72 ( MeCO ), 26.17, 29.00, 29.16, 29.44 and $33.88\left(\mathrm{CH}_{2}\right), 60.35\left(\mathrm{C}-6^{\prime}\right), 62.84(\mathrm{C}-6), 66.50\left(\mathrm{C}-4^{\prime}\right), 68.81$ $\left(\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 69.28(\mathrm{C}-5), 69.47\left(\mathrm{C}-2^{\prime}\right), 70.64$ ( $3 \mathrm{C}, \mathrm{C}-2,-3$ and $\left.-5^{\prime}\right), 71.05$ (C-3'), 74.17 (C-4), 97.62 (C-1), 101.15 (C-1'), 114.23 $\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 128.49-130.03$ and $133.43(\mathrm{Ph}), 139.30\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$, $164.80,165.31$ and $166.05\left(\mathrm{PhCO}_{2}\right), 169.36,169.88$ and 170.16 ( $\mathrm{MeCO}_{2}$ ). Also isolated was dec-9-enyl 2,3,6-tri-O-benzoyl-4-O-(2,3,4,6-tetra-O-acetyl- $\alpha$-D-galactopyranosyl)- $\alpha$-D-mannopyranoside $22(0.14 \mathrm{~g}, 15 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+38$ (c 1, $\mathrm{CHCl}_{3}$ ); $R_{\mathrm{f}} 0.46$ (solvent $B$ ) (Found: C, 63.4; H, $6.4 \%$ ); $\delta_{\mathrm{H}} 1.34$ $\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.67\left(2 \mathrm{H}\right.$, quintet, $\left.J 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right)$, $1.79,1.90,1.95$ and $2.06(12 \mathrm{H}, 4 \times \mathrm{s}, 4 \times \mathrm{Ac}), 2.05(2 \mathrm{H}$, quartet, $\left.J 6.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right), 3.53$ and $3.79\left(2 \mathrm{H}, 2 \times \mathrm{dt},{ }^{2} J_{\mathrm{H} . \mathrm{H}}\right.$ $\left.9.8,{ }^{3} J_{\mathrm{H}, \mathrm{H}} 6.6, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.89\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{a}^{\prime}} 6.0,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.05$ ( $1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime} ; 6 \mathrm{~b}^{\prime}} 10.8,6^{\prime}-\mathrm{H}^{\mathrm{b}}$ ), 4.26 ( 1 H , ddd, $J_{5,6 \mathrm{a}} 4.1,5-\mathrm{H}$ ), $4.36\left(1 \mathrm{H}, \mathrm{ddd}, J_{5^{\prime}, 6 b^{\prime}} 7.1,5^{\prime}-\mathrm{H}\right), 4.54\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.3\right.$, $4-\mathrm{H}), 4.61\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.8,6-\mathrm{H}^{2}\right), 4.86\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 1.7,6-\right.$ $\left.\mathbf{H}^{\mathrm{b}}\right), 4.90-5.06\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}\right.$ and $\left.1-\mathrm{H}\right), 5.16\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime} \cdot 3^{\prime}}\right.$ $\left.10.9,2^{\prime}-\mathrm{H}\right), 5.30\left(1 \mathrm{H}, \mathrm{dd}, J_{3^{\prime}, 4^{\prime}} 3.0,3^{\prime}-\mathrm{H}\right), 5.44\left(1 \mathrm{H}, \mathrm{dd}, J_{4^{\prime}, 5^{\prime}}\right.$ $\left.0.9,4^{\prime}-\mathrm{H}\right), 5.53\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{\prime}} 3.8,1^{\prime}-\mathrm{H}\right), 5.62\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 1.9,2-\right.$ H), 5.74 ( $1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.4,3-\mathrm{H}$ ), 5.83 [ 1 H , ddt, $J_{\mathrm{H}, \mathrm{H}} 10.2, J_{\mathrm{H}, \mathrm{H}}$ $16.9, J\left(\mathrm{H}_{2} \mathrm{CH}_{2}\right) 6.5, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ] and 7.31-8.15 $(15 \mathrm{H}, \mathrm{m}$, $3 \times \mathrm{Ph}) ; \delta_{\mathrm{C}} 20.41$ and $20.62(\mathrm{MeCO}), 26.16,29.01,29.18,29.45$, 29.77 and $33.89\left(\mathrm{CH}_{2}\right), 61.30(\mathrm{C}-6$ ) $), 63.71(\mathrm{C}-6), 66.73\left(\mathrm{C}-2^{\prime}\right)$, 67.48 (C-3'), 67.56 (C-5'), $67.84\left(\mathrm{C}-4^{\prime}\right), 68.91\left(\mathrm{OCH}_{2} \mathrm{CH}_{2}\right)$, 69.12 (C-5), 70.31 (C-2), 71.95 (C-4), 73.23 (C-3), 97.48 (2 C, C1 and $\left.-1^{\prime}\right), 114.23\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 128.61,129.71,129.78,133.40$ and $133.54(\mathrm{Ph}), 139.17\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 165.42$ and $166.25\left(\mathrm{PhCO}_{2}\right)$ and $170.0\left(\mathrm{MeCO}_{2}\right)$.

Dec-9-enyl 2,3,6-tri- $O$-benzoyl-4- $O$-(2,3,4-tri- $O$-benzoyl- $\beta$-D-galactopyranosyl)- $\alpha$-D-mannopyranoside 7
A solution of HCl in MeOH [prepared at $0^{\circ} \mathrm{C}$ from acetyl chloride ( $0.2 \mathrm{~cm}^{3}$ ) and methanol ( $5 \mathrm{~cm}^{3}$ )] was added to a solution of compound $21(0.524 \mathrm{~g}, 0.546 \mathrm{mmol})$ in $\mathrm{CHCl}_{3}$ ( 1 $\mathrm{cm}^{3}$ ), and the mixture was kept at $24^{\circ} \mathrm{C}$ for 20 h ; TLC (solvent $F$ ) then showed the formation of one major product ( $R_{\mathrm{f}} 0.55$; presumed to be the deacetylated compound 23). The mixture was diluted with $\mathrm{CHCl}_{3}\left(200-250 \mathrm{~cm}^{3}\right)$ and the solution was washed successively with saturated aq. $\mathrm{NaHCO}_{3}$ and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was dissolved in pyridine ( $10 \mathrm{~cm}^{3}$ ), $p, p^{\prime}$-dimethoxytriphenylmethyl chloride $(0.222 \mathrm{~g}, 0.655 \mathrm{mmol})$ was added, and the solution was kept at room temperature for 16 h , whereafter a second portion of the reagent ( $0.2 \mathrm{~g}, 0: 59 \mathrm{mmol}$ ) was added. After a further 20 h , benzoyl chloride ( $0.4 \mathrm{~cm}^{3}, 3.44 \mathrm{mmol}$ ) was added at $-10^{\circ} \mathrm{C}$ and the mixture was stirred overnight at $24^{\circ} \mathrm{C}$. Work-up as described above gave the crude product $24\left(R_{\mathrm{f}} 0.64\right.$, solvent $A$ ), which was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(30 \mathrm{~cm}^{3}\right)$, and $3 \%$ TFA in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(30 \mathrm{~cm}^{3}\right)$ was added at $0^{\circ} \mathrm{C}$. After 2 min , the solution was washed in turn with saturated aq. $\mathrm{NaHCO}_{3}$ and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. Column chromatography [hexane-ethyl acetate, (2:1)] gave the benzoylated disaccharide $7(0.426 \mathrm{~g}, 71 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{24}+91.2$ (c 1, $\mathrm{CHCl}_{3}$ ); $R_{\mathrm{f}} 0.15$ (solvent $A$ ), 0.40 (solvent $B$ ) (Found: C, 69.6; $\mathrm{H}, 6.0 . \mathrm{C}_{64} \mathrm{H}_{64} \mathrm{O}_{17}$ requires $\left.\mathrm{C}, 69.6 ; \mathrm{H}, 5.8 \%\right) ; \delta_{\mathrm{H}} 1.30(10 \mathrm{H}, \mathrm{m}$, $\left.5 \times \mathrm{CH}_{2}\right), 1.63\left(2 \mathrm{H}\right.$, quintet, $\left.J 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 2.05(2 \mathrm{H}$, quartet, $J 6.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=$ ), $3.07\left(1 \mathrm{H}\right.$, dd, $\left.J_{5^{\prime}, 6 \mathrm{a}} \cdot 7.2,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right)$, 3.17 ( $1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 12.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}$ ), $3.21(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 3.46$ and $3.72\left(2 \mathrm{H}, 2 \times \mathrm{dt},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 9.5,{ }^{3} J_{\mathrm{H}, \mathrm{H}} 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.61(1 \mathrm{H}$, dd, $\left.J_{5^{\prime}, 6 \mathrm{~b}^{\prime}} 6.0,5^{\prime}-\mathrm{H}\right), 4.16\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5,6 \mathrm{a}} 3.3,5-\mathrm{H}\right), 4.50(1 \mathrm{H}$, dd, $\left.J_{6 \mathrm{a}, 6 \mathrm{~b}} 12.0,6-\mathrm{H}^{\mathrm{a}}\right), 4.55\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.2,4-\mathrm{H}\right), 4.70$ ( $1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 1.5,6-\mathrm{H}^{\mathrm{b}}$ ), 4.89-5.06 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}$ ), 4.96 ( 1 $\left.\mathrm{H}, \mathrm{d}, J_{1,2} 1.5,1-\mathrm{H}\right), 5.00\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2^{2}} 7.9,1^{\prime}-\mathrm{H}\right), 5.44(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{3^{\prime}, 4^{\prime}} 3.3,3^{\prime}-\mathrm{H}\right), 5.62\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{and} 4^{\prime}-\mathrm{H}\right), 5.80\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime}, 3^{\prime}}\right.$ $\left.10.1,2^{\prime}-\mathrm{H}\right), 5.82\left[1 \mathrm{H}, \mathrm{ddt}, J_{\mathrm{H}, \mathrm{H}} 10.2, J_{\mathrm{H}, \mathrm{H}} 16.5, J\left(\mathrm{H}, \mathrm{CH}_{2}\right) 6.5\right.$, $\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ], $5.89\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.6,3-\mathrm{H}\right)$ and $7.10-8.11$ ( 30 $\mathrm{H}, \mathrm{m}, 6 \times \mathrm{Ph})$; $\delta_{\mathrm{C}} 25.87,28.74,28.89,29.16$ and $33.63\left(\mathrm{CH}_{2}\right)$, 59.53 (C-6'), 62.38 (C-6), $68.30\left(\mathrm{C}-4^{\prime}\right), 68.51\left(\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 69.17$ (C-5), 70.21 (C-2'), 70.30 (C-3), 70.72 (C-2), 71.79 (C-3'), 73.84 (2 C, C-4 and -5'), $97.33(\mathrm{C}-1), 101.14\left(\mathrm{C}-1\right.$ '), $113.99\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$, 128.16-129.91 and 132.87-133.63 ( Ph ), $139.05\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$ and 164.94-166.43 ( $\mathrm{PhCO}_{2}$ ).

## 1,3,4,6-Tetra- $O$-acetyl-2- $O$-(2,3,4,6-tetra- $O$-benzoyl- $\alpha$-D-mannopyranosyl)- $\beta$-D-mannopyranose 27

Benzobromomannose $25\left\{[\alpha]_{\mathrm{D}}^{25}+12.8\left(c 1, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}} 4.50(1\right.$ $\left.\mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{a}} 3.5,6-\mathrm{H}^{2}\right), 4.65\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5,6 \mathrm{~b}} 2.0,5-\mathrm{H}\right), 4.74(1 \mathrm{H}$, dd, $J_{6 \mathrm{a} .6 \mathrm{~b}} 12.2,6-\mathrm{H}^{\mathrm{b}}$ ), $5.91\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.0,2-\mathrm{H}\right.$ ), 6.19-6.35 ( 2 $\mathrm{H}, \mathrm{m}, 3-\mathrm{and} 4-\mathrm{H}), 6.59\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.3,1-\mathrm{H}\right)$ and 7.21-8.18 (20 $\mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph})\}$ was prepared from 1,2,3,4,6-penta- $O$-benzoyl$\alpha, \beta$-D-mannopyranose as described in ref. 10 . The reaction of compounds $25(0.604 \mathrm{~g}, 0.92 \mathrm{mmol})$ and $26(0.21 \mathrm{~g}, 0.603 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in the presence of $\mathrm{AgO}_{3} \mathrm{SCF}_{3}(0.273 \mathrm{~g}, 1.06 \mathrm{mmol})$ and $2,4,6$-collidine ( $0.105 \mathrm{~cm}^{3}, 0.8 \mathrm{mmol}$ ) was accomplished at $-20 \rightarrow+20^{\circ} \mathrm{C}$, as described for the preparation of the disaccharide 11. Column chromatography (solvent $C$ ) gave the mannobiose derivative $27(0.559 \mathrm{~g}, 99.9 \%$ ) as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}-44\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.40$ (solvent $C$ ) (Found: C, $62.5 ; \mathrm{H}, 5.0 . \mathrm{C}_{48} \mathrm{H}_{46} \mathrm{O}_{19}$ requires C, 62.2; $\mathrm{H}, 5.0 \%$ ); $\delta_{\mathrm{H}} 2.05$, 2.19, 2.20 and $2.22(12 \mathrm{H}, 4 \times \mathrm{s}, 4 \times \mathrm{Ac}), 3.88\left(1 \mathrm{H}, \mathrm{ddd}, J_{5,6 \mathrm{a}}\right.$ $2.3,5-\mathrm{H}), 4.24\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a} .6 \mathrm{~b}} 12.1,6-\mathrm{H}^{\mathrm{a}}\right), 4.30\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 2.6\right.$, $2-\mathrm{H}), 4.38\left(1 \mathrm{H}, \mathrm{dd}, J_{5.6 \mathrm{~b}} 4.7,6-\mathrm{H}^{\mathrm{b}}\right)$, 4.44 ( 1 H , dd, $J_{5^{\prime}, 6 \mathrm{a}^{\prime}} 3.0,6^{\prime}-$ $\left.\mathrm{H}^{\mathrm{a}}\right), 4.71\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime} ; 6 \mathrm{~b}^{\prime}} 12.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right)$, $4.86\left(1 \mathrm{H}\right.$, ddd, $J_{5^{\prime}, 6 \mathrm{~b}^{\prime}}$ $\left.2.7,5^{\prime}-\mathrm{H}\right), 5.23(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}), 5.32\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime} \cdot 2^{\prime}} 1.7,1^{\prime}-\mathrm{H}\right), 5.50$ $\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.6,4-\mathrm{H}\right), 5.79\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime} \cdot 3}, 2.9,2^{\prime}-\mathrm{H}\right)$, $5.90\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 0.9,1-\mathrm{H}\right), 6.07\left(1 \mathrm{H}, \mathrm{dd}, 3^{\prime}-\mathrm{H}\right), 6.28(1 \mathrm{H}, \mathrm{t}$, $\left.J_{3^{\prime} .4^{\prime}}=J_{4^{\prime} .5^{\prime}}=10.0,4^{\prime}-\mathrm{H}\right)$ and $7.22-8.10(20 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph})$;
$\delta_{\mathrm{C}} 20.75,20.83$ and $21.10(\mathrm{MeCO}), 61.81(\mathrm{C}-6), 62.43\left(\mathrm{C}-6^{\prime}\right)$, 65.85 (C-4), 66.69 (C-4'), 69.27 (2 C, C-3' and $\left.-5^{\prime}\right), 70.71$ (C-2'), 72.08 (C-3), 73.25 (C-5), 74.85 (C-2), 91.03 (C-1), 98.49 (C-1'), $128.40-129.89,133.24$ and $133.60(\mathrm{Ph}), 165.21,165.31,165.64$ and $166.23\left(\mathrm{PhCO}_{2}\right), 168.52,169.44,170.50$ and 171.12 $\left(\mathrm{MeCO}_{2}\right)$.

## 3,4,6-Tri- $O$-acetyl-2- $O$-(2,3,4,6-tetra- $O$-benzoyl- $\alpha$-D-manno-pyranosyl)- $\alpha$-D-mannopyranose 28

Anhydrous dimethylamine $\left(0.2 \mathrm{~cm}^{3}, 3.0 \mathrm{mmol}\right)$ was added to a solution of compound $27(0.508 \mathrm{~g}, 0.549 \mathrm{mmol})$ in MeCN ( 5 $\mathrm{cm}^{3}$ ) at $-20^{\circ} \mathrm{C}$, and the mixture was then kept at $20^{\circ} \mathrm{C}$ for 5 h , with monitoring by TLC (solvent $C$ ). The mixture was concentrated and MeCN was evaporated off from the residue. The residue was then dissolved in $\mathrm{CHCl}_{3}\left(200 \mathrm{~cm}^{3}\right)$; the solution was washed with water $\left(3 \times 70 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated to give the $\alpha$-hydroxy derivative $28(0.485 \mathrm{~g}$, $99.8 \%$ ) as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}-55\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.28$ (solvent $C$ ) (Found: $\mathrm{C}, 62.6 ; \mathrm{H}, 4.9 . \mathrm{C}_{46} \mathrm{H}_{44} \mathrm{O}_{18}$ requires C , $62.4 ; \mathrm{H}, 5.0 \%) ; \delta_{\mathrm{H}} 2.05,2.14$ and $2.19(9 \mathrm{H}, 3 \times \mathrm{s}, 3 \times \mathrm{Ac}), 4.16$ ( $2 \mathrm{H}, \mathrm{m}, 5-\mathrm{and} 6-\mathrm{H}), 4.24\left(1 \mathrm{H}, \mathrm{dd}, J_{2,3} 3.6,2-\mathrm{H}\right), 4.31(1 \mathrm{H}$, dd, $\left.J_{5,6 \mathrm{~b}} 3.2, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.5,6-\mathrm{H}^{\mathrm{b}}\right), 4.43\left(1 \mathrm{H}, \mathrm{dd}, J_{5^{\prime}, 6 \mathrm{a}^{\prime}} 4.0,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.57$ $\left(1 \mathrm{H}\right.$, ddd, $\left.J_{5^{\prime}, 6 \mathrm{~b}^{\prime}} 2.0,5^{\prime}-\mathrm{H}\right), 4.67\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 12.0,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right)$, $5.25\left(1 \mathrm{H}, \mathrm{d}, J_{1^{\prime}, 2}, 1.7,1^{\prime}-\mathrm{H}\right), 5.48\left(3 \mathrm{H}, \mathrm{m}, J_{1,2} 1.9,1-, 3-\right.$ and 4H), $5.75\left(1 \mathrm{H}, \mathrm{dd}, J_{2^{\prime} .3}, 3.3,2^{\prime}-\mathrm{H}\right), 5.99\left(1 \mathrm{H}, \mathrm{dd}, 3^{\prime}-\mathrm{H}\right), 6.14(1$ $\left.\mathbf{H}, \mathbf{t}, J_{3^{\prime}, 4^{\prime}}=J_{4^{\prime}, 5^{\prime}}=9.9,4^{\prime}-\mathrm{H}\right)$ and $7.21-8.10(20 \mathrm{H}, \mathrm{m}$, $4 \times \mathrm{Ph}) ; \delta_{\mathrm{C}} 20.83,20.92$ and $21.58(\mathrm{MeCO}), 62.43(\mathrm{C}-6), 62.76$ (C-6'), $66.50\left(\mathrm{C}-4^{\prime}\right), 67.04(\mathrm{C}-4), 68.48(\mathrm{C}-5), 69.36\left(\mathrm{C}-3^{\prime}\right), 69.51$ (C-5'), $70.05(\mathrm{C}-3), 70.68\left(\mathrm{C}-2^{\prime}\right), 77.45(\mathrm{C}-2), 92.96\left({ }^{1} J_{\mathrm{C}, \mathrm{H}} 171\right.$, $\mathrm{C}-1$ ), 99.24 (C-1'), 128.38-129.93, 133.23 and $133.60(\mathrm{Ph})$, $165.17,165.34,165.72$ and $166.21\left(\mathrm{PhCO}_{2}\right)$ and $169.62,170.68$ and $171.24\left(\mathrm{MeCO}_{2}\right)$.

Triethylammonium 3,4,6-tri- $O$-acetyl-2-O-(2,3,4,6-tetra- $O$ -benzoyl- $\alpha$-D-mannopyranosyl)- $\alpha$-D-mannopyranosyl hydrogenphosphonate 5
This compound was prepared from compound $28(0.221 \mathrm{~g}, 0.25$ mmol ) as described for the derivative 6. Column chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N},(98: 1: 1) \rightarrow(93: 6: 1)\right]$ gave the hydrogenphosphonate $5(0.24 \mathrm{~g}, 92 \%$ ) as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}-22\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.43$ (solvent $\left.F\right) ; \delta_{\mathrm{H}} 1.23(9 \mathrm{H}, \mathrm{t}$, $\left.3 \times \mathrm{MeCH}_{2}\right), 2.08,2.17$ and $2.20(9 \mathrm{H}, 3 \times \mathrm{s}, 3 \times \mathrm{Ac}), 2.93(6$ $\left.\mathrm{H}, \mathrm{q}, 3 \times \mathrm{MeCH}_{2}\right), 4.17\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J_{4,5} 10.0,5-\mathrm{H}\right), 4.30(3 \mathrm{H}$, $\mathrm{m}, 2-\mathrm{H}$ and $\left.6-\mathrm{H}_{2}\right), 4.43\left(1 \mathrm{H}\right.$, dd, $\left.J_{5^{\prime}, 6 \mathrm{a}^{\prime}} 3.0,6^{\prime}-\mathrm{H}^{\mathrm{a}}\right), 4.62(1 \mathrm{H}$, ddd, $\left.J_{5^{\prime}, 6 b^{\prime}} 2.0,5^{\prime}-\mathrm{H}\right), 4.76\left(1 \mathrm{H}\right.$, dd, $\left.J_{6 \mathrm{a}^{\prime}, 6 \mathrm{~b}^{\prime}} 12.2,6^{\prime}-\mathrm{H}^{\mathrm{b}}\right), 5.28(1$ $\left.\mathrm{H}, \mathrm{d}, J_{1^{\prime} .2} \cdot 1.3,1^{\prime}-\mathrm{H}\right), 5.52(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{and} 4-\mathrm{H}), 5.76(1 \mathrm{H}, \mathrm{dd}$, $\left.J_{2^{\prime} .3^{\prime}} 2.9,2^{\prime}-\mathrm{H}\right), 5.87\left(1 \mathrm{H}, \mathrm{dd}, J_{1,2} 1.5, J_{1, \mathrm{P}} 8.9,1-\mathrm{H}\right), 5.99(1 \mathrm{H}$, dd, $\left.3^{\prime}-\mathrm{H}\right), 6.23\left(1 \mathrm{H}, \mathrm{t}, J_{3^{\prime}, 4^{\prime}}=J_{4^{\prime}, 5^{\prime}}=10.0,4^{\prime}-\mathrm{H}\right), 7.01(1 \mathrm{H}, \mathrm{d}$, $\left.J_{\mathrm{H}, \mathrm{P}} 632.6, \mathrm{HP}\right)$ and $7.25-8.14(20 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{Ph}) ; \delta_{\mathrm{C}} 9.34$ and $45.63(\mathrm{Et}), 20.54$ and $20.63(\mathrm{MeCO}), 62.10(\mathrm{C}-6), 62.32\left(\mathrm{C}-6^{\prime}\right)$, 66.19 (C-4), 66.50 (C-4'), 69.39 (C-3'), 69.51 ( $2 \mathrm{C}, \mathrm{C}-5$ and $-5^{\prime}$ ), 70.27 (C-3), 70.54 (C-2'), 77.12 (d, $J_{\mathrm{C}, \mathrm{P}} 5.4, \mathrm{C}-2$ ), 93.34 (d, $J_{\mathrm{C} . \mathrm{P}}$ 3.6, ${ }^{1} J_{\mathrm{C}, \mathrm{H}} 170, \mathrm{C}-1$ ), 98.91 (C-1'), 128.11-129.82 and $132.81-$ $133.30(\mathrm{Ph}), 164.93,165.09,165.37$ and $165.96\left(\mathrm{PhCO}_{2}\right)$ and 169.29, 170.25 and $170.87\left(\mathrm{MeCO}_{2}\right) ; \delta_{\mathbf{P}}$ 1.88; FAB-MS $(+)$ data: $m / z 1050.74\left(100 \%,[\mathrm{M}+\mathrm{H}]^{+}\right)\left(\mathrm{C}_{52} \mathrm{H}_{60} \mathrm{NO}_{20} \mathrm{P}\right.$ requires M , 1049.35).

## Dec-9-enyl 2,3,4-tri- $O$-benzoyl- $\beta$-D-galactopyranosyl-

 $(1 \rightarrow 4)$-2,3,6-tri- $O$-benzoyl- $\alpha$-D-mannopyranoside 6 - $[2,3,4$-tri-$O$-benzoyl- $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha-$ D-mannopyranosyl phosphate], triethylammonium salt 29 A mixture of compounds $6(129 \mathrm{mg}, 0.09 \mathrm{mmol})$ and $7(99 \mathrm{mg}$, 0.09 mmol ) was dried by evaporation of pyridine $\left(3 \times 1 \mathrm{~cm}^{3}\right)$ therefrom. The residue was dissolved in pyridine $\left(1 \mathrm{~cm}^{3}\right)$, adamantane-1-carbonyl chloride $(47.5 \mathrm{mg}, 0.239 \mathrm{mmol})$ was added, and the mixture was stirred at $20^{\circ} \mathrm{C}$ for 30 min , whereafter a freshly prepared solution of iodine ( $46 \mathrm{mg}, 0.18$mmol ) in $95 \%$ aq. pyridine ( $2 \mathrm{~cm}^{3}$ ) was added. After 10 min , $\mathrm{CHCl}_{3}$ was added, and the solution was washed successively with cold $1 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ and cold $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ TEA hydrogen carbonate, dried by filtration through cotton wool, and concentrated. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(6 \mathrm{~cm}^{3}\right)$, and $2 \%$ TFA in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(6 \mathrm{~cm}^{3}\right)$ was added at $0^{\circ} \mathrm{C}$. After 1 min , the solution was washed successively with ice-cold saturated aq. $\mathrm{NaHCO}_{3}$ and $0.5 \mathrm{~mol} \mathrm{dm}^{-3}$ TEA hydrogen carbonate, dried by filtration through cotton wool, and concentrated. Column chromatography $\quad\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}, \quad(98.8: 0.2: 1) \rightarrow\right.$ (95:4:1)] of the residue gave the tetrasaccharide monophosphate derivative $29(163 \mathrm{mg}, 81 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+70.2$ $\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.43$ (solvent $E$ ); $\delta_{\mathrm{H}} 1.13\left(9 \mathrm{H}, \mathrm{t}, 3 \times \mathrm{MeCH}_{2}\right)$, $1.30\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.62(2 \mathrm{H}$, quintet, $J 6.5$, $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right), 2.05\left(2 \mathrm{H}\right.$, quartet, $\left.J 6.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=\right)$, $2.79\left(6 \mathrm{H}\right.$, quartet, $\left.3 \times \mathrm{MeCH}_{2}\right), 3.04\left(1 \mathrm{H}, \mathrm{dd}, J_{6 \mathrm{a}, 6 \mathrm{~b}} 11.3,6-\mathrm{H}^{\mathrm{a}}\right.$, $\left.\mathrm{Gal}^{\prime}\right), 3.18\left(1 \mathrm{H}, \mathrm{dd}, 6-\mathrm{H}^{\mathrm{b}}, \mathrm{Gal}^{\prime}\right), 3.21\left(1 \mathrm{H}, \mathrm{dt}, J_{5,6 \mathrm{a}} 9.0, J_{6 \mathrm{a}, 6 \mathrm{~b}}=\right.$ $\left.J_{6 \mathrm{a}, \mathrm{P}}=10.5,6-\mathrm{H}^{\mathrm{a}}, \mathrm{Gal}\right), 3.42$ and $3.66\left(2 \mathrm{H}, 2 \times \mathrm{dt},{ }^{2} J_{\mathrm{H}, \mathrm{H}} 9.5\right.$, $\left.{ }^{3} J_{\mathrm{H}, \mathrm{H}} 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.52\left(1 \mathrm{H}, \mathrm{t}, J_{5,6 \mathrm{a}}=J_{5,6 \mathrm{~b}}=6.3,5-\mathrm{H}\right.$, $\left.\mathrm{Gal}^{\prime}\right), 3.88\left(1 \mathrm{H}\right.$, ddd, $\left.J_{6 \mathrm{~b}, \mathbf{p}} 7.5,6-\mathrm{H}^{\mathrm{b}}, \mathrm{Gal}\right), 4.02\left(1 \mathrm{H}, \mathrm{dt}, J_{5,6 \mathrm{a}}=\right.$ $J_{5,6 \mathrm{~b}}=2.5,5-\mathrm{H}$, Man $), 4.13\left(1 \mathrm{H}, \mathrm{dd}, J_{5,6 \mathrm{~b}} 5.5,5-\mathrm{H}, \mathrm{Gal}\right), 4.33$ (1 H, br d, 5-H, Man'), $4.37\left(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}^{\mathrm{a}}\right.$, Man), $4.46(1 \mathrm{H}, \mathrm{m}$, $6-\mathrm{H}^{\mathrm{b}}$, Man $), 4.48\left(1 \mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.5,4-\mathrm{H}\right.$, Man $), 4.53(1$ $\left.\mathrm{H}, \mathrm{t}, J_{3,4}=J_{4,5}=9.5,4-\mathrm{H}, \mathrm{Man}^{\prime}\right), 4.56\left(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}_{2}\right.$, Man' $\left.^{\prime}\right)$, 4.88-5.40 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}$ ), $4.89\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J}_{1,2} 7.7,1-\mathrm{H}, \mathrm{Gal}^{\prime}\right)$, $4.94\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 2.0,1-\mathrm{H}, \mathrm{Man}\right), 4.97\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 7.5,1-\mathrm{H}, \mathrm{Gal}\right)$, $5.34\left(1 \mathrm{H}, \mathrm{dd}, J_{3.4} 3.5,3-\mathrm{H}, \mathrm{Gal}^{\prime}\right), 5.39\left(1 \mathrm{H}, \mathrm{dd}, J_{3.4} 3.4,3-\mathrm{H}\right.$, Gal), 5.45 ( $1 \mathrm{H}, \mathrm{dd}, J_{1,2} 1.9, J_{1, \mathrm{p}} 7.0,1-\mathrm{H}$, Man' $^{\prime}$ ), $5.55(1 \mathrm{H}$, dd, $J_{2,3} 3.5,2-\mathrm{H}$, Man), $5.57\left(1 \mathrm{H}, \mathrm{d}, 4-\mathrm{H}\right.$, Gal' $\left.^{\prime}\right), 5.67(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}$, Man'), $5.70\left(1 \mathrm{H}\right.$, dd, $\left.J_{2,3} 10.0,2-\mathrm{H}, \mathrm{Gal}^{\prime}\right), 5.71(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}=\mathrm{CH}_{2}\right), 5.77\left(1 \mathrm{H}, \mathrm{dd}, \mathrm{J}_{2,3} 2.5,3-\mathrm{H}, \mathrm{Man}^{\prime}\right), 5.83(2 \mathrm{H}$, dd and d, $J_{2,3} 10.5,2-$ and $\left.4-\mathrm{H}, \mathrm{Gal}\right), 5.89(1 \mathrm{H}, \mathrm{dd}, 3-\mathrm{H}, \mathrm{Man})$ and $7.10-$ $8.09(60 \mathrm{H}, \mathrm{m}, 12 \times \mathrm{Ph}) ; \delta_{\mathrm{C}} 9.49$ and $45.68(\mathrm{Et}), 25.95,28.82$, $29.00,29.24$ and $33.72\left(\mathrm{CH}_{2}\right), 60.04\left(\mathrm{C}-6, \mathrm{Gal}^{\prime}\right), 61.49\left(\mathrm{~d}, J_{\mathrm{C}, \mathrm{P}}\right.$ 5.1, C-6, Gal), 62.26 (C-6, Man), 62.51 (C-6, Man'), 67.16 (C-4, $\mathrm{Gal})$, 68.44 (2 C, $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2}+\mathrm{C}-4, \mathrm{Gal}^{\prime}\right), 69.50$ (C-3, Man'), 69.77 (C-5, Man), 69.83 (C-3, Man), 69.86 (C-5, Man'), 70.20 (C-2, Gal'), 70.40 (C-2, Gal), 70.70 (d, $\left.J_{\mathrm{C}, \mathrm{P}} 8.5, \mathrm{C}-2, \mathrm{Man}^{\prime}\right), 71.17$ (C-2, Man), 71.87 (C-3, Gal'), 72.0 (d, $\left.J_{\mathrm{C}, \mathrm{P}} 7.3, \mathrm{C}-5, \mathrm{Gal}\right), 72.50$ (C-3, Gal), 72.83 (C-4, Man'), 74.08 (C-4, Man), 74.14 (C-5, Gal'), 93.44 (d, $\left.J_{\mathrm{C}, \mathrm{P}} 4.8, \mathrm{C}-1, \mathrm{Man}^{\prime}\right), 97.28$ (C-1, Man), 100.45 (C-1, Gal'), 101.46 (C-1, Gal), $114.04\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 128.20$ 129.60, 132.94 and $133.17(\mathrm{Ph}), 139.13\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$ and $164.93-$ $166.45\left(\mathrm{PhCO}_{2}\right) ; \delta_{\mathbf{P}}-2.95\left(\mathrm{dt}, J_{\mathbf{P}, \mathrm{H}} 7.5\right.$ and 10.5$) ;$ FAB-MS( + ) data: $m / z 2157.77\left(70 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}+\mathrm{Na}\right]^{+}\right)$and 2236.22 $\left(100,[\mathrm{M}+\mathrm{H}]^{+}\right)\left(\mathrm{C}_{124} \mathrm{H}_{124} \mathrm{NO}_{36} \mathrm{P}\right.$ requires $\left.\mathrm{M}, 2235.31\right)$.

The condensation of compounds 6 and 7 in pyridine in the presence of either trimethylacetyl chloride ( 2.5 mol equiv.) or bis-(2-oxooxazolidin-3-yl)phosphinic chloride ( 2.5 mol equiv.), followed by oxidation and detritylation as described above, gave the compound 29 in 79 or $70 \%$ yield, respectively.

Dec-9-enyl 2,3,4-tri- $O$-benzoyl- $\beta$-D-galactopyranosyl-(1 $\rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-D-mannopyranoside 6 -\{2,3,4-tri- $O$ -benzoyl- $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-Dmannopyranosyl phosphate 6 -[ $2,3,4$-tri- $O$-benzoyl- $\beta$-D-galacto-pyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-D-mannopyranosyl phosphate] $\}$, bistriethylammonium salt 30
This compound was prepared by condensation of the disaccharide synthon $6(165 \mathrm{mg}, 0.115 \mathrm{mmol})$ and the tetrasaccharide block $29(183 \mathrm{mg}, 0.082 \mathrm{mmol})$ in the presence of adamantane-1-carbonyl chloride ( $57 \mathrm{mg}, 0.288 \mathrm{mmol}$ ), followed by oxidation with iodine ( $46 \mathrm{mg}, 0.18 \mathrm{mmol}$ ) and treatment with $0.7 \% \mathrm{TFA}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(1 \mathrm{~min}, 0^{\circ} \mathrm{C}\right)$, as described in the preparation of the compound 29 . Column chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N},(98.7: 0.3: 1) \rightarrow(96: 3: 1)\right]$ gave the hexasaccharide bisphosphate derivative $30(207 \mathrm{mg}, 75 \%)$ as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+64.5\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.32$
(solvent $E$ ); $\delta_{\mathrm{c}} 9.82$ and 45.91 (Et), 26.10, 28.80-29.20 and 33.61 ( $\mathrm{CH}_{2}$ ), 59.98 (C-6, Gal"), 61.36 (2 C, d, $J_{\mathrm{C}, \mathrm{P}} 4.2, \mathrm{C}-6, \mathrm{Gal}+\mathrm{C}-$ 6, Gal'), 62.06, 62.26 and 62.39 (C-6, Man; C-6, Man'; C-6, Man"), 67.10 and 67.11 (C-4, Gal; C-4, Gal'), 68.38 (2 C, $\left.\mathrm{OCH}_{2} \mathrm{CH}_{2}+\mathrm{C}-4, \mathrm{Gal}^{\prime \prime}\right), 69.43$ (2 C, C-3, Man' $+\mathrm{C}-3$, Man"), 69.73 ( 2 C ), 69.77 (C-3 and -5, Man; C-5, Man"), 70.13 (C-5, Man'), 70.18 and 70.20 (C-2, Gal'; C-2, Gal"), 70.39 (C-2, Gal), 70.57 and 70.76 ( $2 \mathrm{~d}, J_{\mathrm{C}, \mathrm{p}} 7.4$ and 6.3, C-2, Man'; C-2, Man"), 71.14 (C-2, Man), 71.83 (C-3, Gal"), 71.90 and 72.0 ( $2 \mathrm{~d}, J_{\mathrm{C}, \mathrm{P}} 7.6$ and 7.1, C-5, Gal; C-5, Gal'), 72.38 and 72.43 (C-3, Gal; C-3, $\mathrm{Gal}^{\prime}$ ), 72.76 (C-4, Man"), 73.39 (C-4, Man'), 73.99 (C-4, Man), 74.07 (C-5, Gal"), 93.33 ( $2 \mathrm{C}, \mathrm{d}, J_{\mathrm{C} . \mathrm{P}} 4.2, \mathrm{C}-1, \mathrm{Man}^{\prime}+\mathrm{C}-1$, Man"), 97.23 (C-1, Man), 100.33 (C-1, Gal"), 101.19 (C-1, $\left.\mathrm{Gal}^{\prime}\right), 101.40(\mathrm{C}-1, \mathrm{Gal}), 114.01\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 128.40-129.82$ and 132.90-133.90 ( Ph ), $139.45\left(\mathrm{CH}=\mathrm{CH}_{2}\right)$ and 164.65-166.09 $\left(\mathrm{PhCO}_{2}\right) ; \delta_{\mathrm{P}}-2.99\left(\mathrm{dt}, J_{\mathrm{P}, \mathrm{H}} 7.5\right.$ and $\left.10.5, \mathrm{P}\right)$ and $-3.08\left(\mathrm{dt}, J_{\mathrm{P}, \mathrm{H}}\right.$ 7.5 and $\left.10.5, \mathrm{P}^{\prime}\right)$ ) ES-MS(-) data: $m / z 1580.1(100 \%$, [M - 2 $\left.\left.\mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}\right]^{2-}\right)\left(\mathrm{C}_{184} \mathrm{H}_{184} \mathrm{~N}_{2} \mathrm{O}_{55} \mathrm{P}_{2}\right.$ requires $\left.\mathrm{M}, 3363.12\right)$.

## Dec-9-enyl 2,3,4-tri- $O$-benzoyl- $\beta$-d-galactopyranosyl-(1 $\rightarrow$ 4)-2,3,6-tri- $O$-benzoyl- $\alpha$-D-mannopyranoside 6 - [ $2,3,4$-tri- $O$ -benzoyl- $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-Dmannopyranosyl phosphate 6 - $\{2,3,4$-tri- $O$-benzoyl- $\beta$-D-galacto-pyranosyl-( $1 \rightarrow 4$ )-2,3,6-tri- $O$-benzoyl- $\alpha$-D-mannopyranosyl phosphate 6-[2,3,4,6-tetra- O-benzoyl- $\alpha$-D-mannopyranosyl( $\mathbf{1} \boldsymbol{\rightarrow 2}$ )-3,4,6-tri-O-acetyl- $\alpha$-D-mannopyranosyl phosphate]\}], tristriethylammonium salt 31

This compound was prepared by condensation of the disaccharide synthon $5(52.5 \mathrm{mg}, 0.05 \mathrm{mmol})$ and the hexasaccharide block $30(107 \mathrm{mg}, 0.0318 \mathrm{mmol})$ in the presence of adamantane- 1 -carbonyl chloride ( $25 \mathrm{mg}, 0.125 \mathrm{mmol}$ ), followed by oxidation with iodine ( $25 \mathrm{mg}, 0.1 \mathrm{mmol}$ ), as described for the synthesis of compound 29. Column chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right.$-water- $\mathrm{Et}_{3} \mathrm{~N}$, (98.8:0.18: $0.02: 1) \rightarrow(89: 9: 1: 1)]$ gave the protected octasaccharide trisphosphate 31 ( $125 \mathrm{mg}, 89 \%$ ) as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}$ $+31.2\left(c 1, \mathrm{CHCl}_{3}\right) ; R_{\mathrm{f}} 0.60$ (solvent $G$ ); $\delta_{\mathrm{C}} 8.44$ and 45.51 (Et), 25.92, 28.84-29.52, $33.61\left(\mathrm{CH}_{2}\right), 61.38$ ( $3 \mathrm{C}, \mathrm{br}, \mathrm{C}-6$, $\left.\mathrm{Gal}+\mathrm{C}-6, \mathrm{Gal}^{\prime}+\mathrm{C}-6, \mathrm{Gal}^{\prime}\right), 61.93,62.07,62.14,62.32$ and 62.43 (C-6, Man; C-6, Man'; C-6, Man"; C-6, Man'"; C-6, Man""), 65.92 (C-4, Man"'), 66.53 (C-4, Man""), 67.14 (3 C, C$\left.4, \mathrm{Gal}+\mathrm{C}-4, \mathrm{Gal}^{\prime}+\mathrm{C}-4, \mathrm{Gal}^{\prime \prime}\right), 68.42\left(\mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 69.07(\mathrm{C}-$ 5, Man"), 69.27 (C-3, Man"'), 69.44 ( $2 \mathrm{C}, \mathrm{C}-3$, Man' + C-3, Man"), 69.72 (2 C), 69.81 (C-3 and -5, Man; C-5, Man"'"), 70.14 (2 C, C-5, Man' + C-5, Man"), 70.20 and 70.24 (C-2, Gal'; C-2, Gal"), 70.31 (C-3, Man"'), 70.43 (C-2, Gal), 70.51 and 70.82 ( 2 d, J J, $\mathbf{P}$ 7.2, C-2, Man'; C-2, Man"), 70.61 (C-2, Man""), 71.19 (C-2, Man), 71.61 (d, J $\mathrm{C}_{\mathrm{C}}$ 7.2, C-5, Gal"), 71.85 and 71.95 ( 2 d , $\left.J_{\mathrm{C}, \mathrm{P}} 7.2, \mathrm{C}-5, \mathrm{Gal} ; \mathrm{C}-5, \mathrm{Gal}^{\prime}\right), 72.37,72.40$ and 72.49 (C-3, Gal; $\mathrm{C}-3, \mathrm{Gal}^{\prime}$; C-3, Gal"), 73.36 and 73.44 (C-4, Man'; C-4, Man"), 74.05 (C-4, Man), 77.21 (d, J. ${ }_{\text {C.P }} 6.0, \mathrm{C}-2$, Man"' $^{\prime}$ ), 93.35 (2 C, br, C-1, Man ${ }^{\prime}$ + C-1, Man"), 94.20 (C-1, br, Man"'), 97.27 (C-1, Man), 98.91 (C-1, Man""), 101.16 (C-1, Gal"), 101.20 (C-1, $\left.\mathrm{Gal}^{\prime}\right)$, $101.44(\mathrm{C}-1, \mathrm{Gal}), 114.02\left(\mathrm{CH}=\mathrm{CH}_{2}\right), 127.82-130.0$ and 132.38-133.27(Ph), 139.12 ( $\left.\mathrm{CH}=\mathrm{CH}_{2}\right), 164.76-166.12\left(\mathrm{PhCO}_{2}\right)$, and $169.05,170.34$ and $170.88\left(\mathrm{MeCO}_{2}\right) ; \delta_{\mathrm{P}}-2.55\left(\mathrm{dt}, J_{\mathrm{P}, \mathrm{H}} 7.5\right.$ and $\left.10.5, \mathrm{P}^{\prime \prime}\right),-2.95\left(\mathrm{dt}, J_{\mathrm{P}, \mathrm{H}} 7.5\right.$ and $\left.10.5, \mathrm{P}\right)$ and $-3.02(\mathrm{dt}$, $J_{\mathrm{P}, \mathrm{H}} 7.5$ and $\left.10.5, \mathrm{P}^{\prime}\right)$; ES-MS(-) data: $\mathrm{m} / \mathrm{z} 1368.5(100 \%$, $\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-3 \mathrm{H}\right]^{3-}$ ) and 2053.6 ( $35,\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-2\right.$ $\left.\mathrm{H}]^{2-}\right)\left(\mathrm{C}_{236} \mathrm{H}_{242} \mathrm{~N}_{3} \mathrm{O}_{75} \mathrm{P}_{3}\right.$ requires $\left.\mathrm{M}, 4410.44\right)$.

## Dec-9-enyl $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyrano-

 side 1To a solution of compound $21(102 \mathrm{mg})$ in $\mathrm{MeOH}\left(20 \mathrm{~cm}^{3}\right)$ was added $4.6 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{NaOMe}$ in $\mathrm{MeOH}\left(0.44 \mathrm{~cm}^{3}\right)$. The mixture was kept at room temp. for 2 h , deionised with Dowex $50 \mathrm{~W}-\mathrm{X} 4$ $\left(\mathrm{H}^{+}\right)$resin, and concentrated to dryness. Water ( $5 \times 10 \mathrm{~cm}^{3}$ ) was evaporated off from the residue to remove methyl
benzoate. The disaccharide 1 ( $51 \mathrm{mg}, 99.99 \%$ ) was thereby obtained as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}+41.5(c 1, \mathrm{MeOH}) ; R_{\mathrm{f}}$ 0.69 (solvent $I$ ); $\delta_{\mathrm{H}}\left(\mathrm{D}_{2} \mathrm{O}\right)$ (inter alia) $1.32\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right)$, $1.60\left(2 \mathrm{H}\right.$, quintet, $J 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ), $2.03(2 \mathrm{H}$, quartet, $J$ $6.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=$ ), $3.32\left(1 \mathrm{H}, \mathrm{dt}, J 6.5, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right), 3.61(1 \mathrm{H}$, dd, $\left.2^{\prime}-\mathrm{H}\right), 3.78\left(3 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right.$ and $\left.3^{\prime}-\mathrm{and} 4-\mathrm{H}\right), 3.89(1 \mathrm{H}$, dd, $3-\mathrm{H}), 3.95\left(1 \mathrm{H}, \mathrm{d}, 4^{\prime}-\mathrm{H}\right), 3.99(1 \mathrm{H}, \mathrm{dd}, 2-\mathrm{H}), 4.48(1 \mathrm{H}, \mathrm{d}$, $\left.J_{1^{\prime}, 2}, 7.5,1^{\prime}-\mathrm{H}\right), 4.82\left(1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.5,1-\mathrm{H}\right), 4.91(1 \mathrm{H}$, br d, $J$ $10.0, \mathrm{CH}=\mathrm{CH}_{2}$ ), $4.97\left(1 \mathrm{H}\right.$, br d, $J 17.0, \mathrm{CH}=\mathrm{CH}_{2}$ ) and 5.77 [ 1 H , ddt, $J\left(\mathrm{H}_{\mathrm{C}} \mathrm{CH}_{2}\right) 6.5, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}$ ]; $\delta_{\mathrm{C}}$, see Table 1; ES-MS ( + ) data: $m / z 503.3\left(100 \%,[\mathrm{M}+\mathrm{Na}]^{+}\right)$and $983.5(22,[2 \mathrm{M}+$ $\mathrm{Na}]^{+}$); ES-MS $(-)$data: $m / z 479.4\left(100 \%,[\mathrm{M}-\mathrm{H}]^{-}\right)$and $959.5\left(15,\left[2 \mathrm{M}-\mathrm{H}^{-}\right)\left(\mathrm{C}_{22} \mathrm{H}_{40} \mathrm{O}_{11}\right.\right.$ requires $\left.\mathrm{M}, 480.26\right)$.

## Dec-9-enyl $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranoside $6^{\text {Ga1 }}$ - $\beta$ - D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranosyl <br> phosphate], triethylammonium salt 2

To a solution of compound $29(70 \mathrm{mg})$ in $\mathrm{MeOH}\left(20 \mathrm{~cm}^{3}\right)$ was added $4.6 \mathrm{~mol} \mathrm{dm}^{-3} \mathrm{NaOMe}$ in $\mathrm{MeOH}\left(0.22 \mathrm{~cm}^{3}\right)$. The mixture was kept for 2 h at $24^{\circ} \mathrm{C}$ and for 16 h at $1^{\circ} \mathrm{C}$, then was deionised with Dowex $50 \mathrm{~W}-\mathrm{X} 4\left(\mathrm{H}^{+}\right)$resin, filtered, and immediately neutralised with $\mathrm{Et}_{3} \mathrm{~N}$. After concentration, water ( $5 \times 10 \mathrm{~cm}^{3}$ ) was evaporated off from the residue to remove methyl benzoate. The tetrasaccharide monophosphate 2 ( 30.5 $\mathrm{mg}, 98.8 \%$ ) was thereby obtained as an amorphous solid; $[\alpha]_{\mathrm{D}}^{22}$ $+40.5(c 1, \mathrm{MeOH}) ; R_{\mathrm{f}} 0.59$ (solvent $H$ ), 0.24 (solvent $\eta$;; $\delta_{\mathrm{P}}\left(\mathrm{D}_{2} \mathrm{O}\right)-1.28 ; \delta_{\mathrm{C}}$, see Table 1; ES-MS(-) data: $m / z 883.4$ $\left(100 \%,\left[\mathrm{M}-\mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}^{-}\right)\right.$and $1789.6\left(2,\left[2 \mathrm{M}-2 \mathrm{Et}_{3} \mathrm{~N}-\right.\right.$ $\left.2 \mathrm{H}+\mathrm{Na}]^{-}\right)\left(\mathrm{C}_{40} \mathrm{H}_{76} \mathrm{NO}_{24} \mathrm{P}\right.$ requires $\left.\mathrm{M}, 985.45\right)$.

Dec-9-enyl $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranoside $6^{\text {Gal- }}\{\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranosyl phosphate $6^{\text {Gal- }}$ [ $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranosyl phosphate]\}, bistriethylammonium salt 3
O-Deacylation of compound $30(55 \mathrm{mg})$ with $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ NaOMe in MeOH , followed by work-up as in the preceding experiment, gave the hexasaccharide diphosphate 3 ( 24 mg , $98.5 \%$ ) as an amorphous solid; $[\alpha]_{\mathrm{D}}^{21}+20(c 1, \mathrm{MeOH}) ; R_{\mathrm{f}} 0.48$ (solvent $H$ ); $\delta_{\mathrm{P}}\left(\mathrm{D}_{2} \mathrm{O}\right)-1.29 ; \delta_{\mathrm{c}}$, see Table 1; ES-MS $(-)$ data: $m / z 643.3\left(100 \%\right.$, $\left.\left[\mathrm{M}-2 \mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}\right]^{2-}\right)$, 693.4 ( 20 , $[\mathrm{M}-$ $\left.\left.\mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}\right]^{2-}\right), 1287.5\left(2,\left[\mathrm{M}-2 \mathrm{Et}_{3} \mathrm{~N}-\mathrm{H}^{-}\right)\right.$) and $1309.4\left(5,\left[\mathrm{M}-2 \mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}+\mathrm{Na}\right]^{-}\right)\left(\mathrm{C}_{58} \mathrm{H}_{112} \mathrm{~N}_{2} \mathrm{O}_{37} \mathrm{P}_{2}\right.$ requires $\mathrm{M}, 1490.64$ ).

## Dec-9-enyl $\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranoside $6^{\text {Gal-[ }}[\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranosyl

 phosphate $6^{\text {Gal- }}\{\beta$-D-galactopyranosyl-( $1 \rightarrow 4$ )- $\alpha$-D-mannopyranosyl phosphate $6^{\text {Gal- }}[\alpha-\mathrm{D}-$ mannopyranosyl- $(1 \rightarrow 2)-\alpha-\mathrm{D}-$ mannopyranosyl phosphate]\}], tristriethylammonium salt 4 O-Deacylation of compound $31(100 \mathrm{mg})$ with $0.05 \mathrm{~mol} \mathrm{dm}^{-3}$ NaOMe in MeOH , followed by work-up as described in the preparation of the tetrasaccharide 2, gave the octasaccharide trisphosphate 4 ( $45 \mathrm{mg}, 99.5 \%$ ) as an amorphous solid; $[\alpha]_{\mathrm{D}}^{21}$ $+39.5(c 1, \mathrm{MeOH}) ; R_{\mathrm{f}} 0.30$ (solvent $H$ ); $\delta_{\mathbf{H}}\left(\mathrm{D}_{\mathbf{2}} \mathrm{O}\right)$ (inter alia) $1.27\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 1.59(2 \mathrm{H}$, quintet, $J 6.5$, $\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ ), 2.05 ( 2 H , quartet, $J 6.5, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}=$ ), 4.46 ( $3 \mathrm{H}, \mathrm{d}, J_{1.2} 7.4,1-\mathrm{H}, \mathrm{Gal}+1-\mathrm{H}, \mathrm{Gal}^{\prime}+1-\mathrm{H}, \mathrm{Gal}^{\prime}$ ), 4.87 ( 1 H , d, $\left.J_{1,2} 1.2,1-\mathrm{H}, \mathrm{Man}\right), 4.96\left(1 \mathrm{H}\right.$, br d, $\left.J 10.0, \mathrm{CH}=\mathrm{CH}_{2}\right), 5.04$ ( 1 H, br d, $J 17.0, \mathrm{CH}=\mathrm{CH}_{2}$ ), 5.06 ( $1 \mathrm{H}, \mathrm{d}, J_{1,2} 1.2,1-\mathrm{H}, \mathrm{Man}{ }^{\prime \prime}$ ), 5.43 ( 2 H , dd, $J_{1.2} 1.2, J_{1 . \mathrm{P}} 7.2,1-\mathrm{H}$, Man' $^{\prime}+1-\mathrm{H}$, Man" $^{\prime \prime}$ ), 5.65 ( 1 H , dd, $J_{1,2} 1.2, J_{1 . \mathrm{P}} 7.2,1-\mathrm{H}, \mathrm{Man}^{\prime \prime}$ ) and $5.92[1 \mathrm{H}$, ddt, $\left.J_{\left(\mathrm{H} . \mathrm{CH}_{2}\right)} 6.5, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right] ; \delta_{\mathrm{P}}\left(\mathrm{D}_{2} \mathrm{O}\right)-1.27 ; \delta_{\mathrm{C}}$, see Table 1; ES-MS(-) data: $m / z 563.3\left(100 \%\right.$, $\left.\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-3 \mathrm{H}\right]^{3-}\right)$, 845.4 (13, $\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}\right]^{2-}$ ), 856.2 ( $40,[\mathrm{M}-3$ $\left.\left.\mathrm{Et}_{3} \mathrm{~N}-3 \mathrm{H}+\mathrm{Na}\right]^{2-}\right), 864.3\left(4,\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-3 \mathrm{H}+\right.\right.$ $\mathrm{K}^{2-}$ ), $867.5\left(4,\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-4 \mathrm{H}+2 \mathrm{Na}\right]^{2-}\right), 1729.5(1$, $\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-2 \mathrm{H}+\mathrm{K}^{-}\right)$and $1735.5\left(2,\left[\mathrm{M}-3 \mathrm{Et}_{3} \mathrm{~N}-\right.\right.$ $\left.3 \mathrm{H}+2 \mathrm{Na}]^{-}\right)\left(\mathrm{C}_{76} \mathrm{H}_{148} \mathrm{~N}_{3} \mathrm{O}_{50} \mathrm{P}_{3}\right.$ requires $\left.\mathrm{M}, 1995.83\right)$.
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